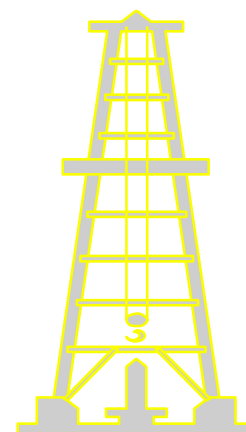


Analysis of Refinery and Petrochemical Products by Capillary GC



System Optimization

- GC theory (**painful, but necessary**)
- Phase selectivity “tuning”
- High-speed GC
- Method Translation Software (**free on Agilent’s website!**)
- Retention Time Locking



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Table of Boiling Point Fractions

Carbon #	C ₁	C ₂	C ₃	C ₄	C ₅	C ₆	C ₇	C ₈	C ₉	C ₁₀	C ₁₁	C ₁₂	C ₁₃	C ₁₄	C ₁₅	C ₁₆
Bpt of n-Paraffin @ 760 mm Hg																
Centigrade	-161	-89	-42	-0.5	+36	69	98	126	151	174	196	216	235	253	270	287
Fahrenheit	-259	-127	-44	+31	97	156	209	258	303	345	384	421	421	488	519	548

Refinery off gas	↔															
Naphtha solvents					↔											
Cracking Naphtha						↔										
Reformate						↔										
Gasoline			↔													
Kerosine, Diesel								↔								to C ₁₈
Jet Fuel								↔								to C ₁₇
Gas Oil									↔							to >C ₂₀
Lube Oil												C ₂₀	↔			
Pitch Asphalt												C ₂₀	↔			
Wax													C ₁₇	↔		
Residues														C ₄₄₊	↔	



Are These *Really* Tools for Improving Chromatography?

$$\beta = \frac{r_c}{2d_f}$$

$$K_c = \frac{W_{i(s)} / V_s}{W_{i(M)} / V_M}$$

$$\alpha = \frac{k_2}{k_1}$$

$$k = \frac{t'_R}{t_m}$$

$$R_s = \frac{\sqrt{N}}{4} \left(\frac{k}{k+1} \right) \left(\frac{\alpha - 1}{\alpha} \right)$$



Chromatography is *Just Two Things*

$$K_c = \frac{\text{Conc. solute in stationary phase}}{\text{Conc. solute in mobile phase}}$$

$$R_s = \frac{\sqrt{N}}{4} \left(\frac{k}{k+1} \right) \left(\frac{\alpha-1}{\alpha} \right)$$



Distribution Constant (K_c)



$$K_c = \frac{\text{conc. of solute in stationary phase}}{\text{conc. of solute in mobile phase}}$$



K_c Describes a Solute's Location

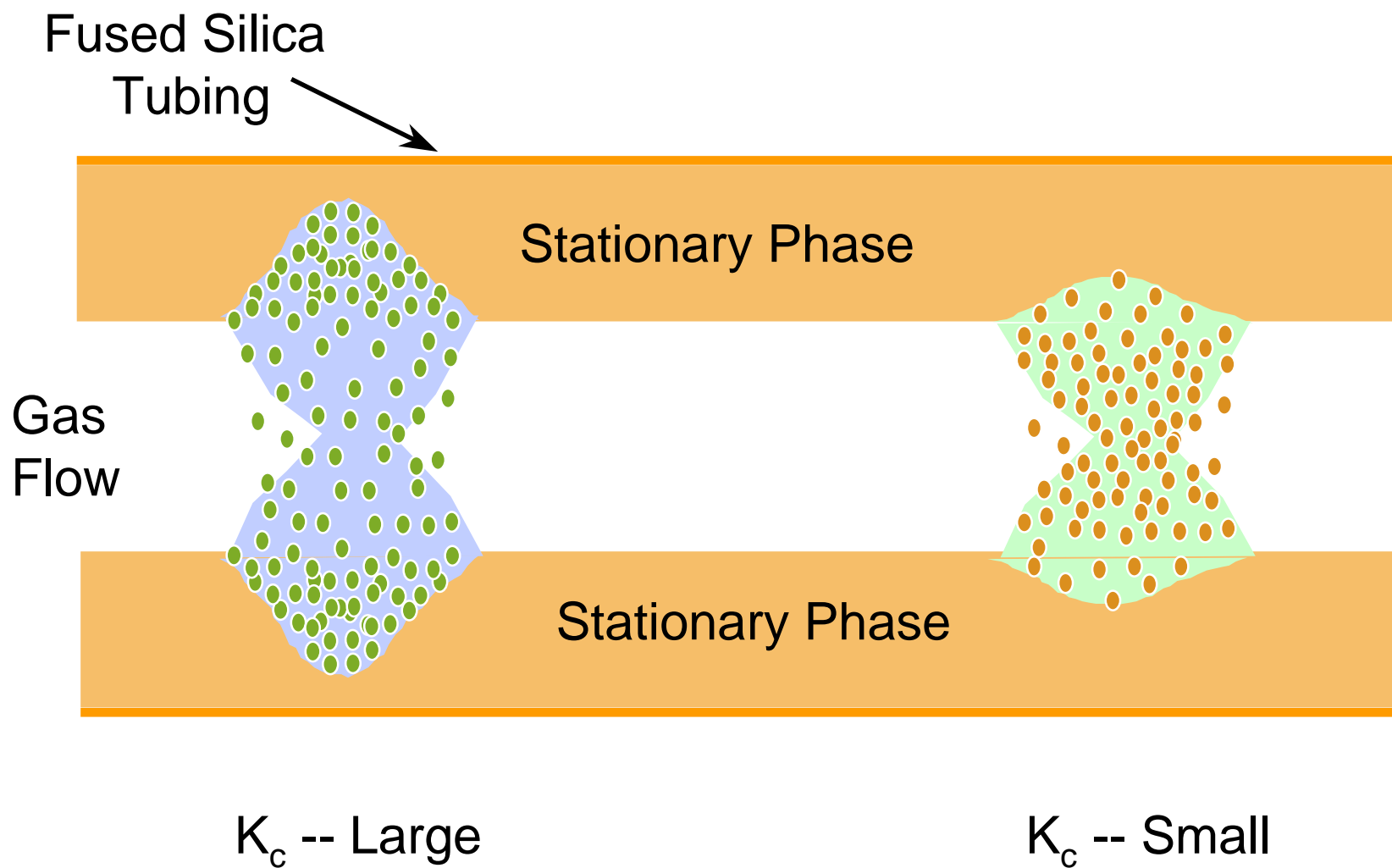
Solute in stationary phase -- Not moving

Solute in mobile phase -- Moving towards detector



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K_c Is Dynamic



Manipulation of Retention

$$K_c = k\beta$$

Distribution Constant

$$k = K_c / \beta$$

Retention Factor

$$\beta = K_c / k$$

Phase Ratio

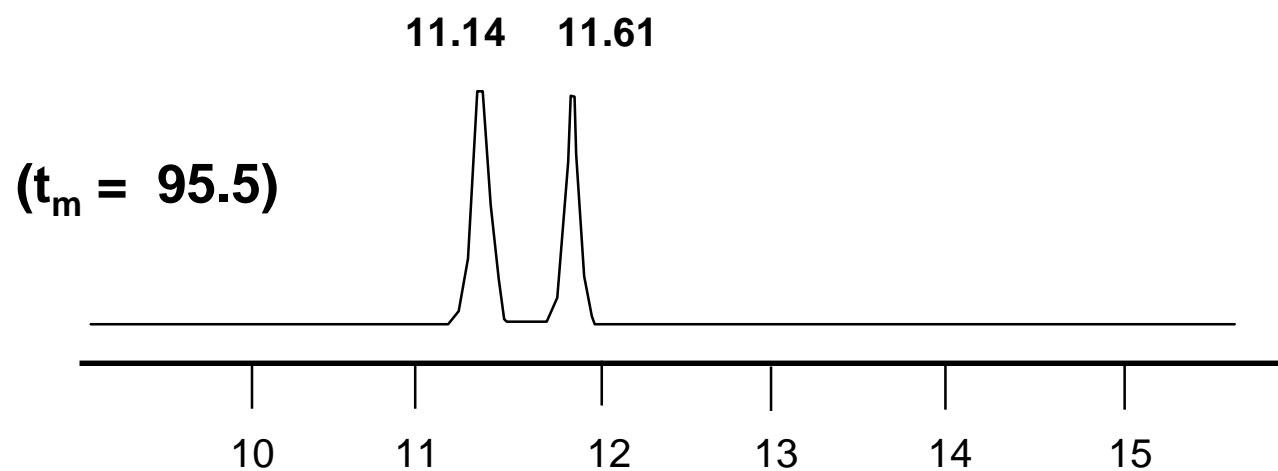
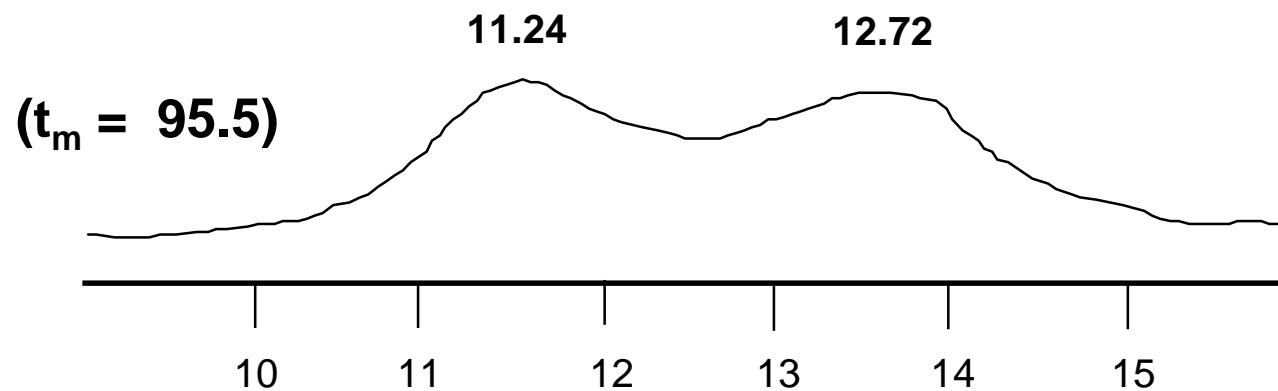


Resolution vs Separation

- Separation: Time between the 2 peaks
- Resolution: Describes how well 2 peaks are separated with respect to their widths



Which pair of solutes have better separation?

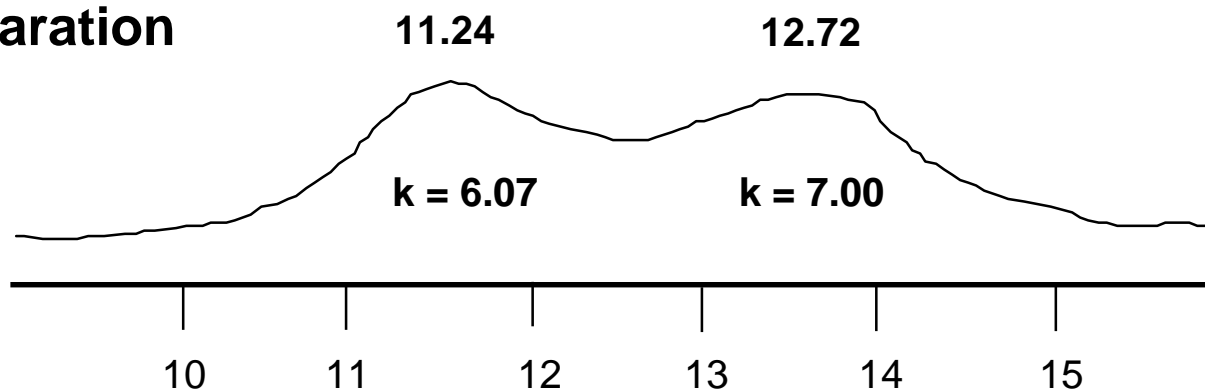


Resolution vs Separation

Better Separation

$$\alpha = 1.17$$

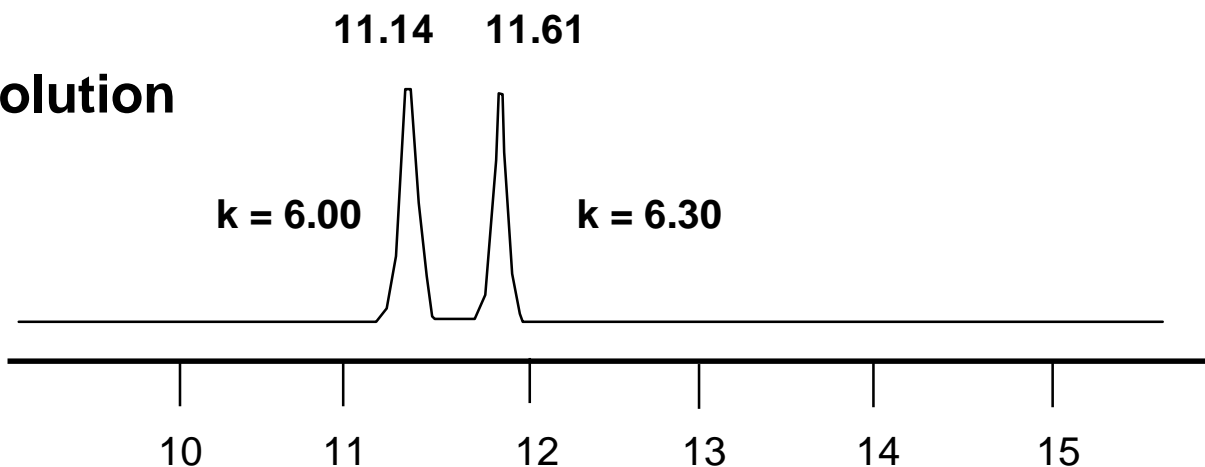
$$R_s = 0.6$$



Better Resolution

$$\alpha = 1.05$$

$$R_s = 2.7$$



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Resolution

$$R_s = \frac{\sqrt{N}}{4} \left(\frac{k}{k+1} \right) \left(\frac{\alpha-1}{\alpha} \right)$$

$$N = f(L, r_c)$$

$$k = f(T, d_f, r_c)$$

$$\alpha = f(T, \text{phase})$$



Detailed Analysis of Gasoline

- Well over 400 components
- Solute differences are subtle
- Intended to identify the majority of HC types in a sample



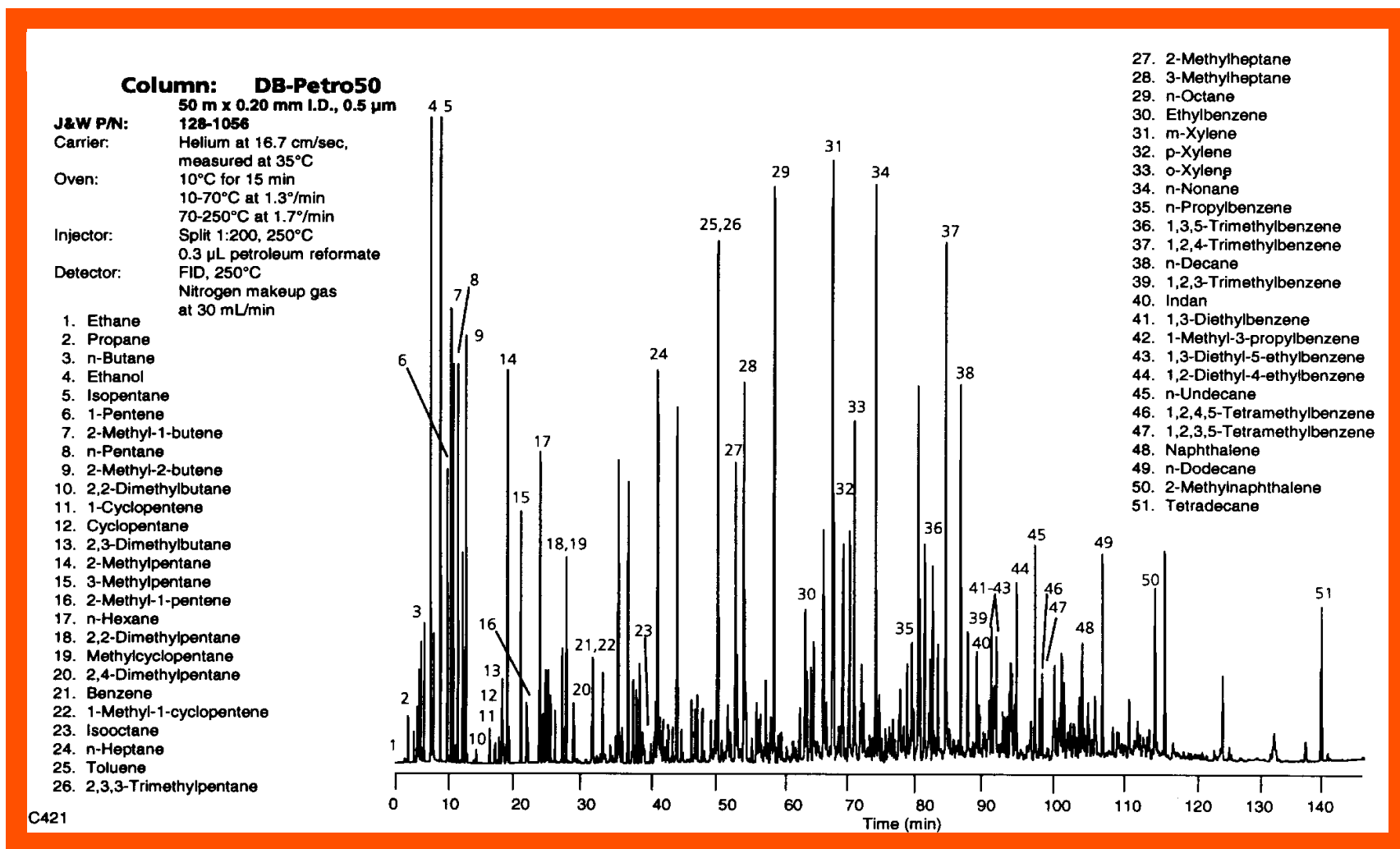
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Methods for Refinery Stream Characterization by Hydrocarbon Type Analysis

- ASTM 5134: Detailed Analysis of Petroleum Naphthas through n-Nonane by CGC
- AFNOR NF M07-086: Determination of HC Group Type in Motor Gasolines from Detailed Analysis by CGC
- CAN/ C.G.S.B.: 3.0, no. - 14.3 - 94: Detailed Analysis of Gasoline
- ASTM D6730
- PONA-BAMA: Paraffins, Olefins, Naphthenes and Aromatics - By Any Means Available



Detailed Analysis of Gasoline by AFNOR Method #2



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"Single" Methyl Silicone Column for Hydrocarbon Type Analysis

- $N > 400,000$
- Oven programming finesse
- Can miss certain key resolutions



Key Solute Resolutions *

Desired in Reformulated Fuel

- MTBE - 2,3-Dimethylbutane
- Benzene - 1-Methylcyclopentane
- Toluene - 2,3,3-Trimethylpentane
- m,p-Xylene - 2,3-Dimethylheptane

* Not readily achieved on methyl siloxane columns



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Detailed Hydrocarbon Analysis (DHA)-ASTM Method D6730-01, tuned column

- Applicable to liquid HC mixtures
- Includes analysis of reformulated gasolines
- Intent is to identify the majority of HC types in a sample



DHA Method General Guidelines

- GC/FID with split injection
- Oven programming from 5 to 200°C (0.1°/min)
- Capillary column: 100 m x 0.25 mm I.D., 0.5 µm methyl siloxane meeting minimum performance requirements
- "Tuning" the methylsiloxane column to achieve key separations



“Tuning” the Methylsiloxane Column To Achieve Key DHA Separations

- Step 1: Add a length of (5% phenyl)-methylsiloxane pre-column for aromatic selectivity
- Step 2: Reduce length of pre-column to achieve all key separations
- Step 3: Adjust oven temperature program if needed



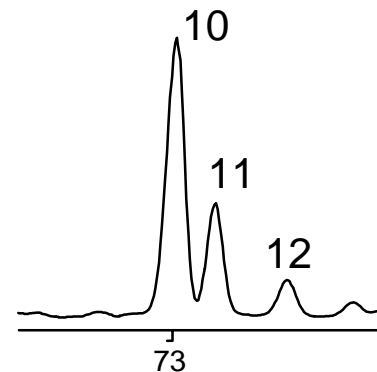
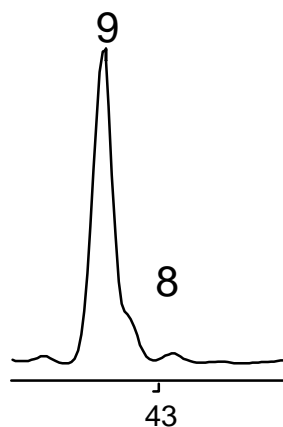
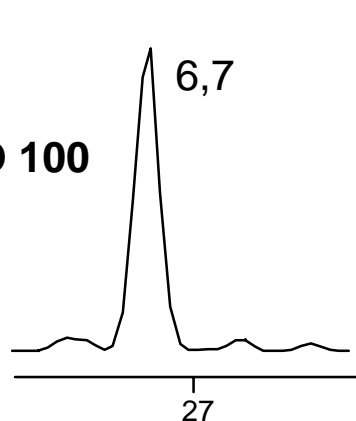
Compound List - critical resolutions

- | | |
|---------------------------|------------------------------|
| 6. 1-Methylcyclopentane | 12. 2,3-Dimethylheptane |
| 7. Benzene | 13. Unidentified isoparaffin |
| 8. 2,2,3-Trimethylpentane | 14. 1,2-Methylethylbenzene |
| 9. Toluene | 15. 2-Methylnaphthalene |
| 10. m-Xylene | 16. 1-Methylnaphthalene |
| 11. p-Xylene | 17. Tridecane |

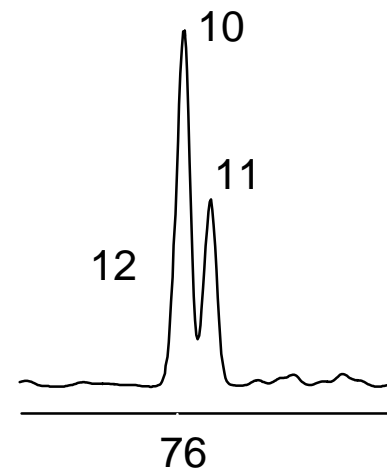
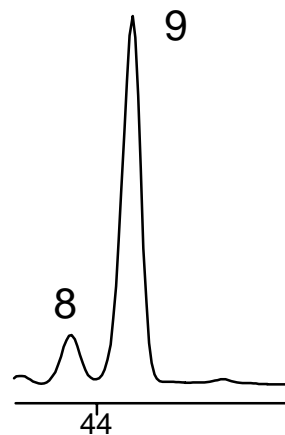
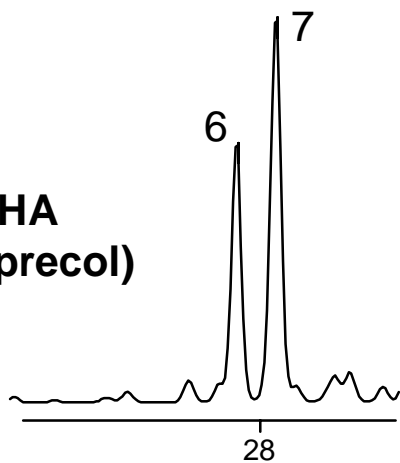


STEP 1 - Adding 5m DB-5 Pre-column

DB-PETRO 100



**HP-DHA
(5 m precol)**

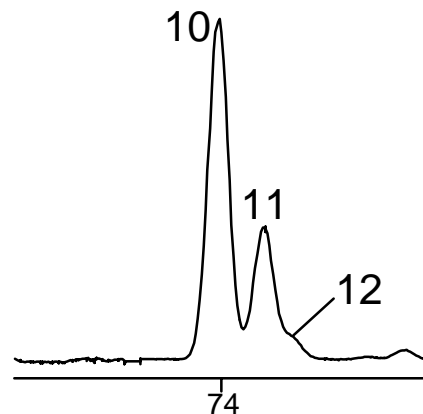
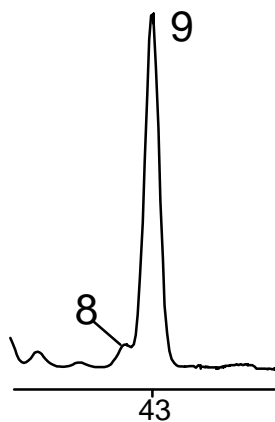
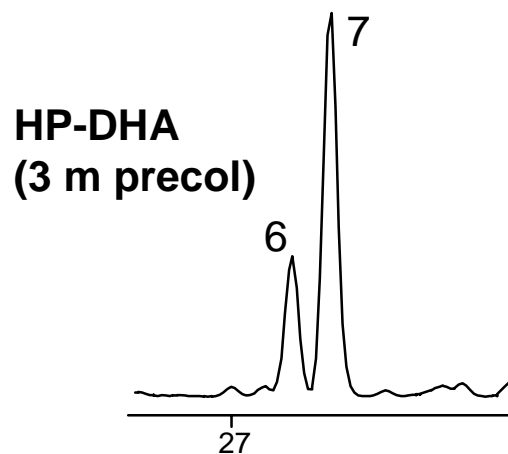
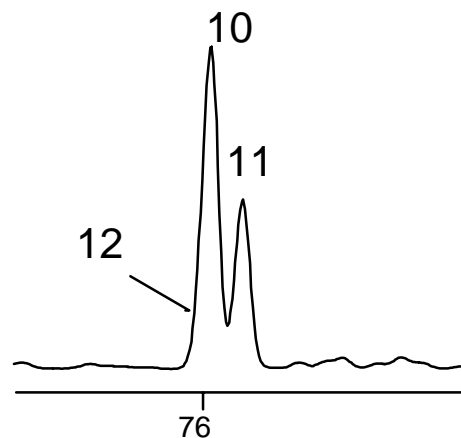
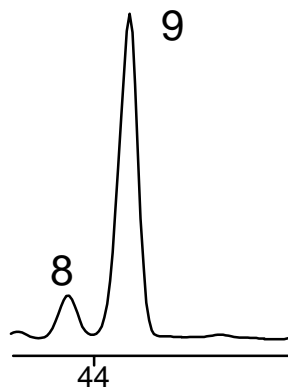
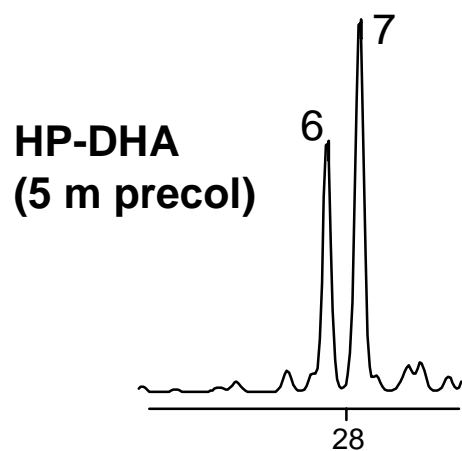


Oven Program: 5°C// 10 min// 5°/min// 50°C// 50 min// 1.5°/min // 200°C



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STEP 2 - Tuning HP-DHA Selectivity

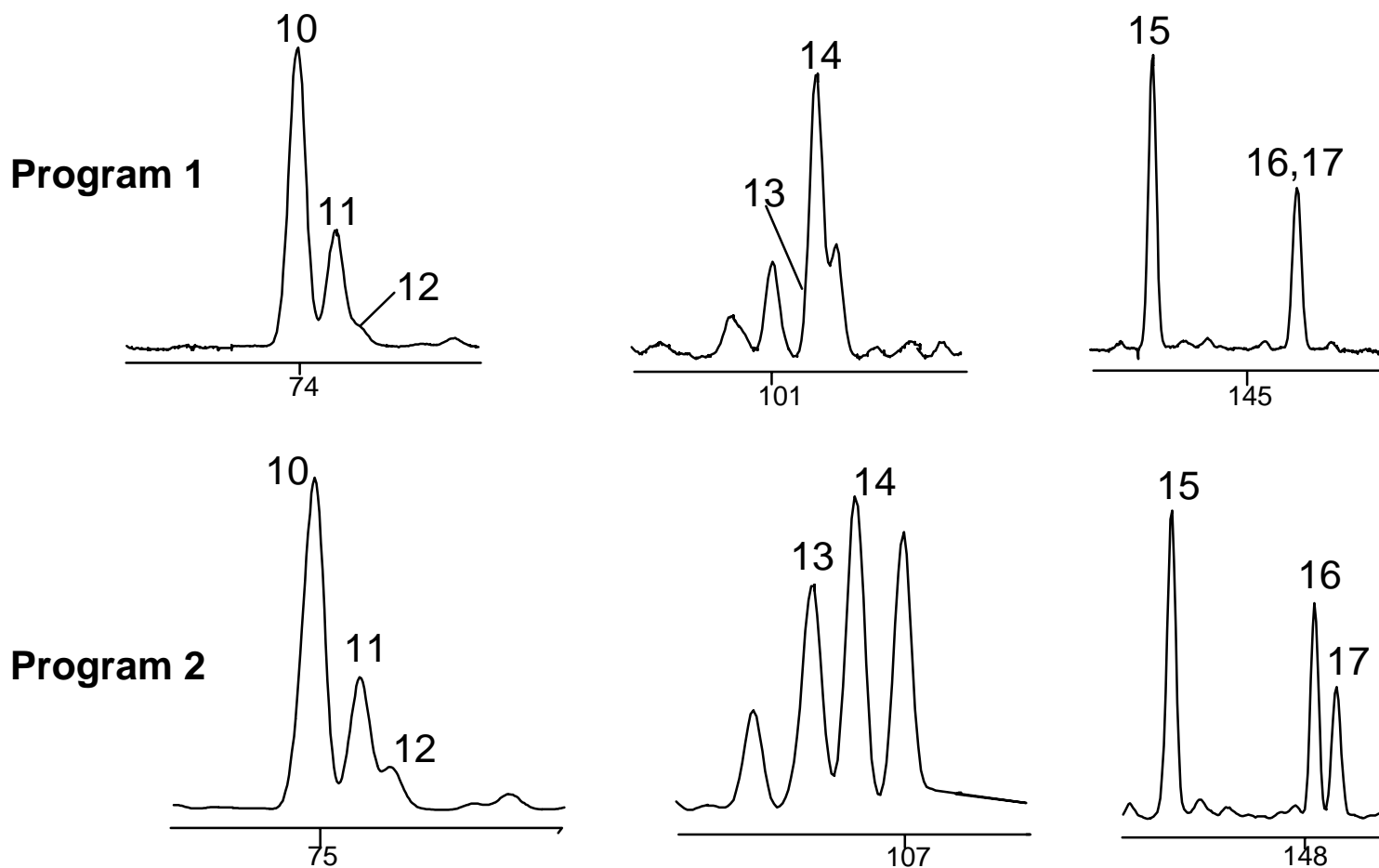


Oven Program: 5°C// 10 min// 5°/min// 50°C// 50 min// 1.5°/min // 200°C



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STEP 3 - Tuning Oven Temperature for "3-meter - DHA"



Oven Program 1: 5°C// 10 min// 5°/min// 50°C// 50 min// 1.5°/min// 200°C

Oven Program 2: 5°C// 10 min// 5°/min// 50°C// 59 min// 1.7°/min// 200°C



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Column: HP-DHA

100 m x 0.25 mm I.D., 0.5 μ m DB-1

5 meters of DB-5, 0.25 mm I.D., 1.0 μ m

Carrier: Helium at 24 cm/sec (measured at 35°C)

Oven: 5°C for 10 min

5-50°C at 5°/min

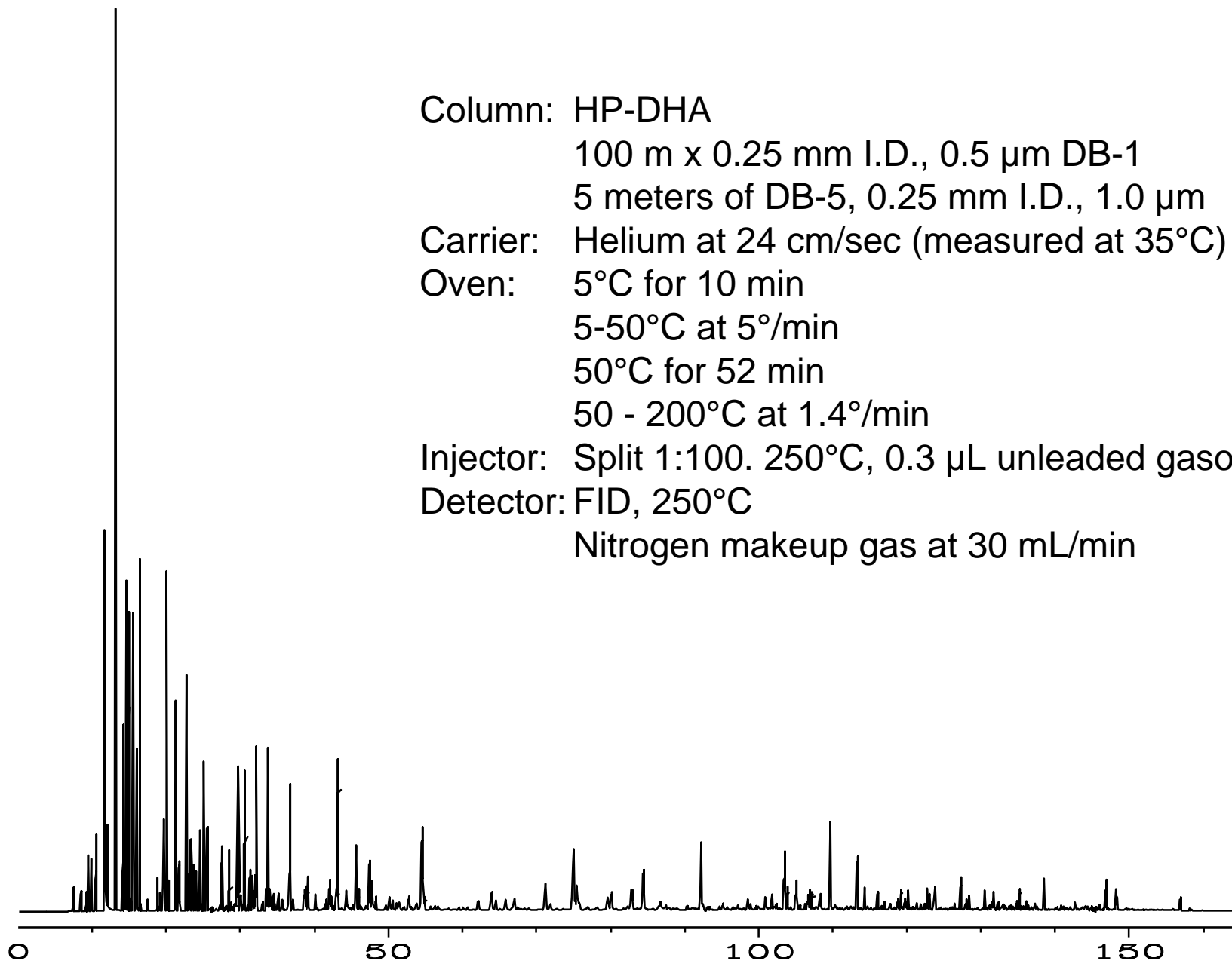
50°C for 52 min

50 - 200°C at 1.4°/min

Injector: Split 1:100. 250°C, 0.3 μ L unleaded gasoline

Detector: FID, 250°C

Nitrogen makeup gas at 30 mL/min



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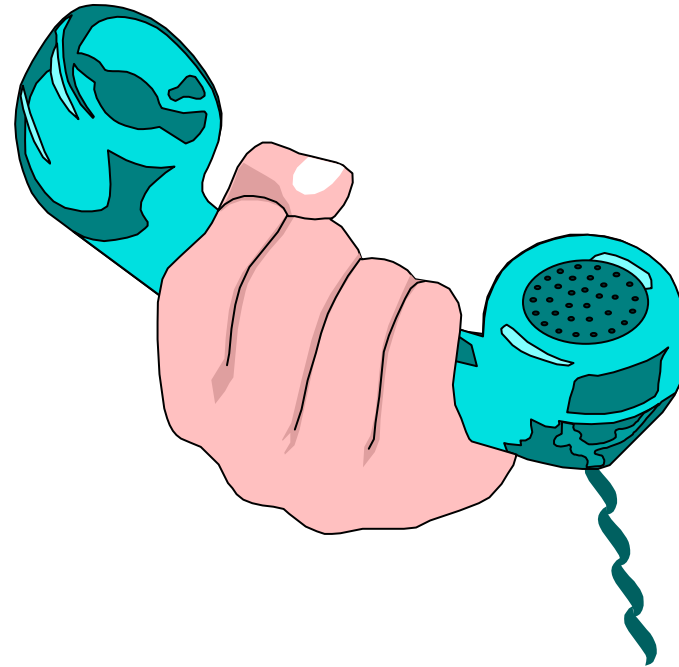
Key Considerations for HC Type Analysis

- What is your sample stream?
- What are your final product specifications?
- Capillary columns with a high N are essential but may not be enough.



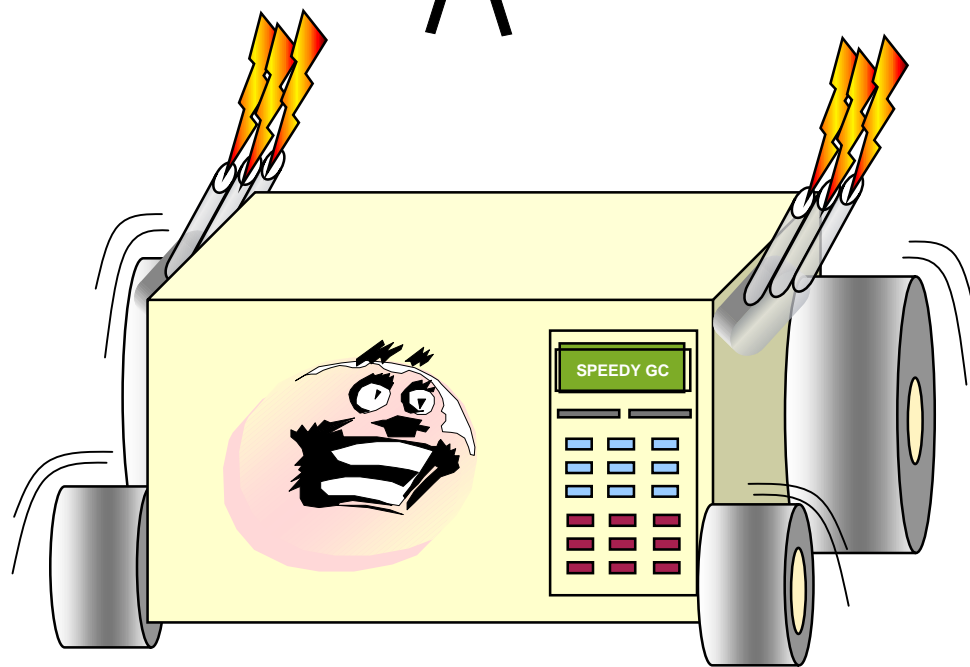
Break Number 1

For Questions and Answers
Press 1 on Your Phone to
Ask a Question



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Speeding Things Up



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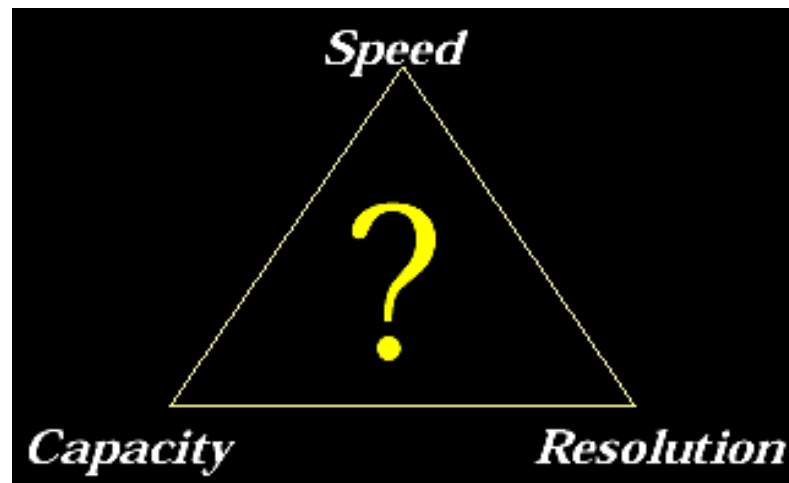
Questions to Ask

- What information do you need from your analysis?
- Do you have more baseline than needed between your peaks?
- Do you need to resolve all of the components?

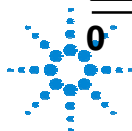
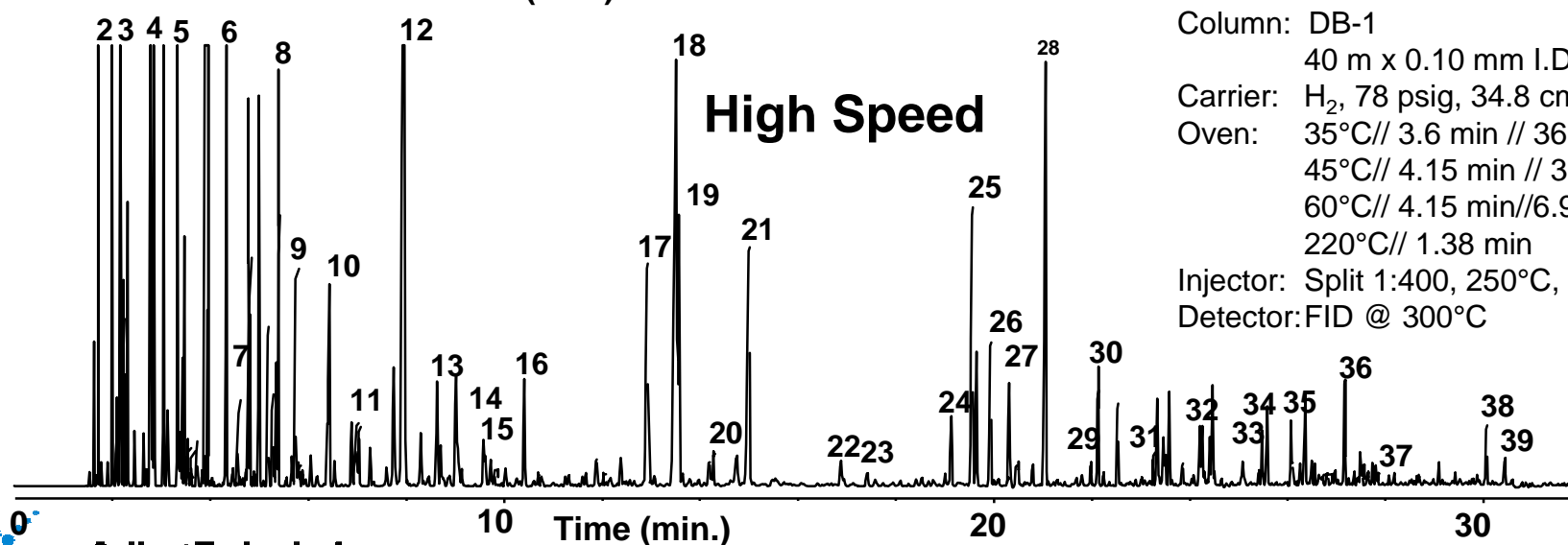
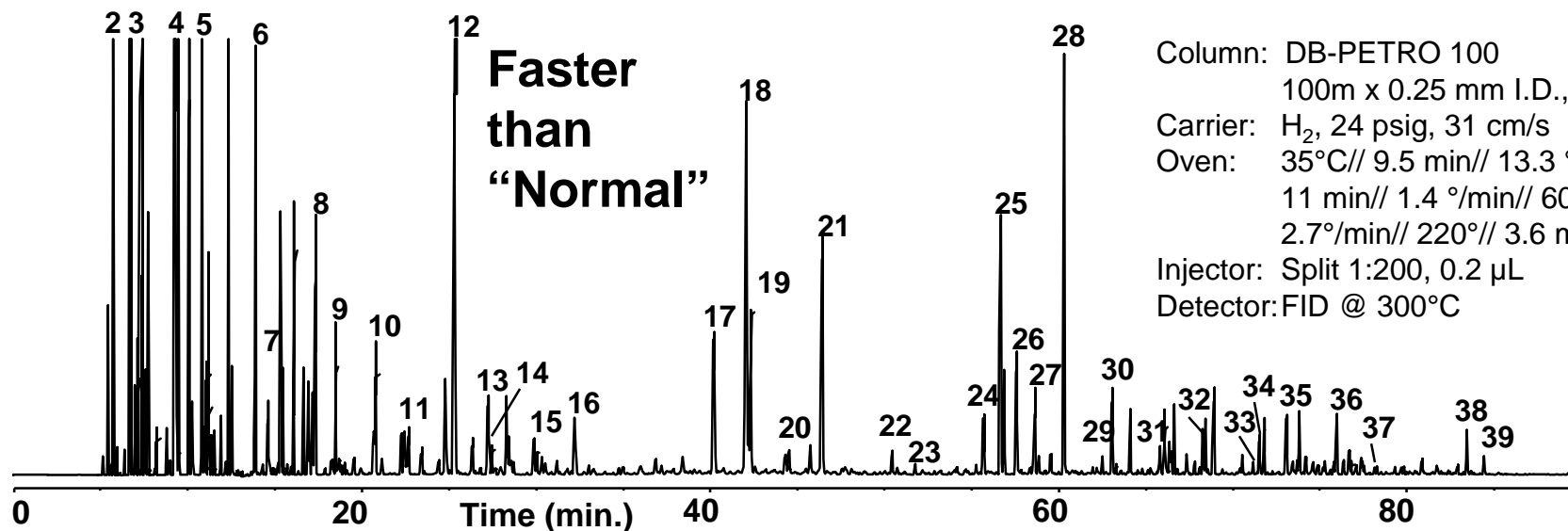


Variables for Shortening Run Times

- Stationary Phase
- Temperature Programming
- Carrier Gas: type and linear velocity
- Shorten Column Length
- Decrease Film Thickness
- Decrease Internal Diameter

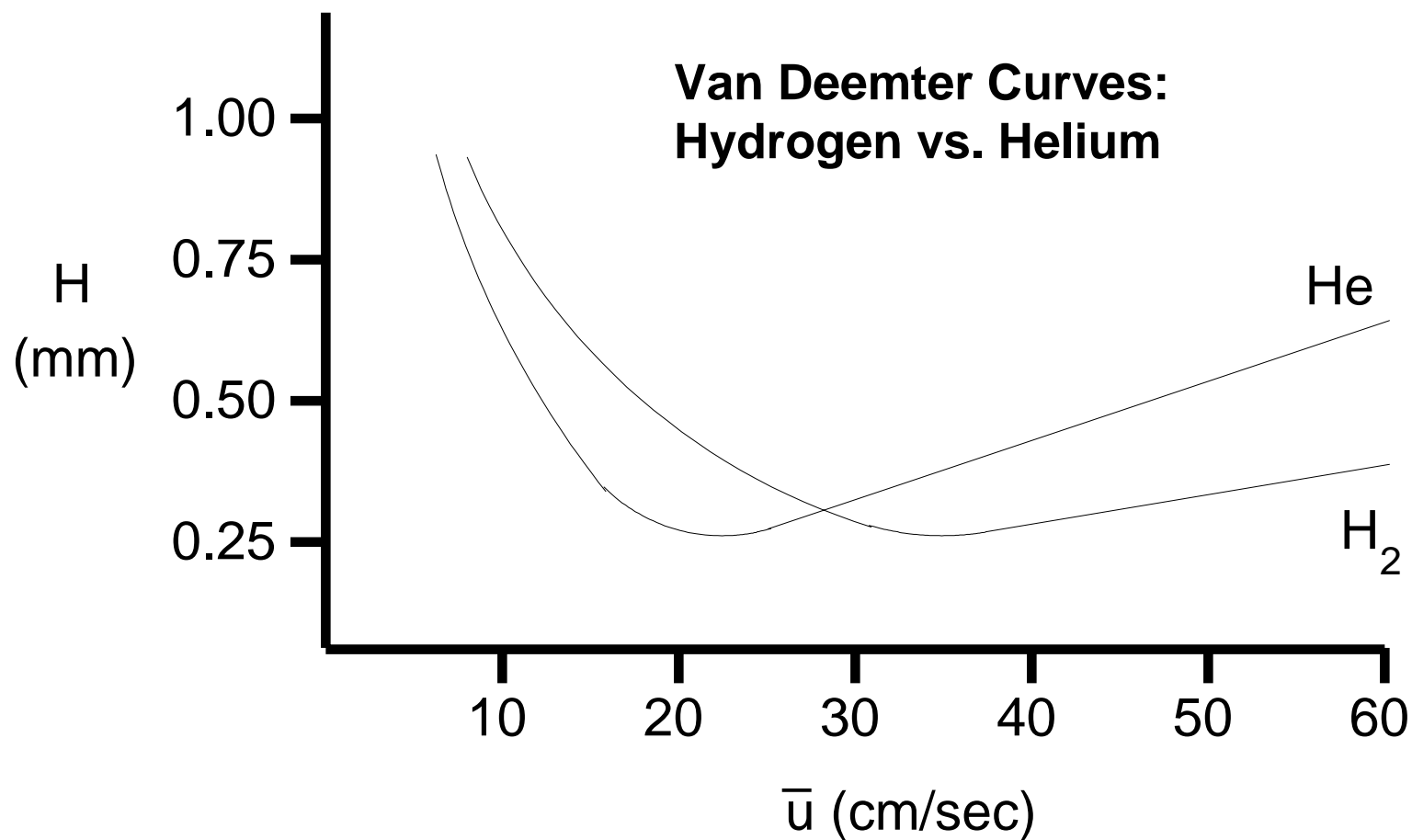


Regular Unleaded California Phase I



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Carrier Gas Type



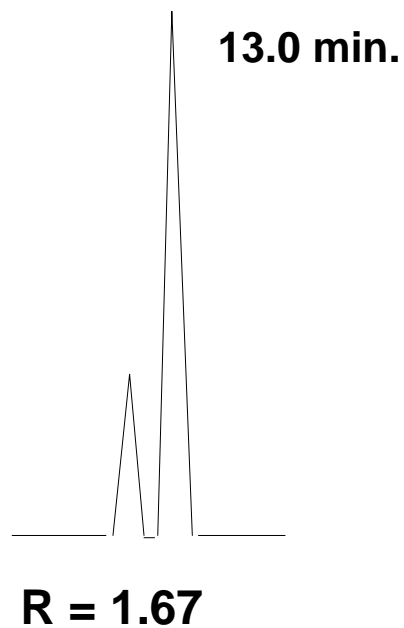
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Hydrogen vs. Helium

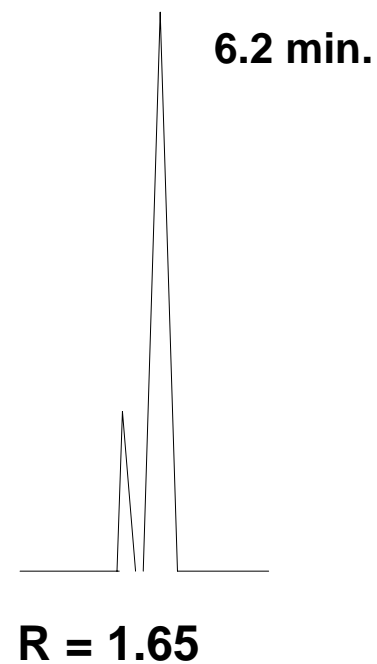
SE-52
15m x 0.25mm
150°C isothermal

Compounds:
C17
Pristane

Helium
23.2 cm/sec



Hydrogen
48 cm/sec



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Factors Affecting Resolution

$$R_s = [(N)^{1/2}/4] [k/(k+1)] [(\alpha-1)/\alpha]$$

Efficiency: N = theoretical plates

Retention: k = retention factor

Selectivity: α = separation factor



Distribution Constant

$$K_c = \frac{\text{conc. of solute in stationary phase}}{\text{conc. of solute in mobile phase}}$$

Change in K_c affects retention

Co-elution if solute K_c 's are equal

K_c is determined by:

solute

stationary phase

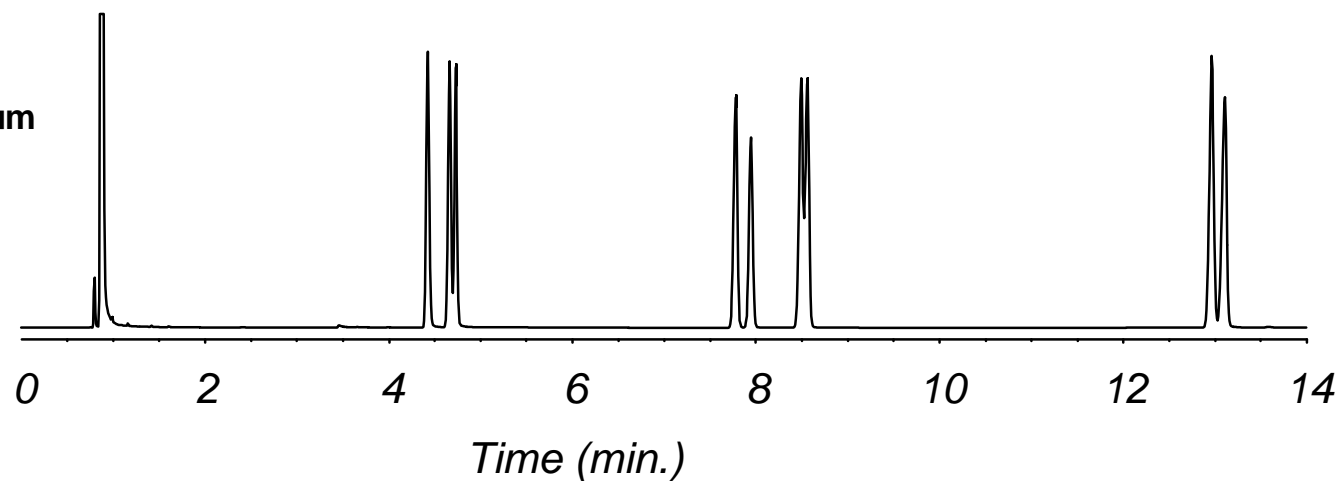
temperature

Stationary phase and temperature changes do not affect the K_c of all solutes equally.

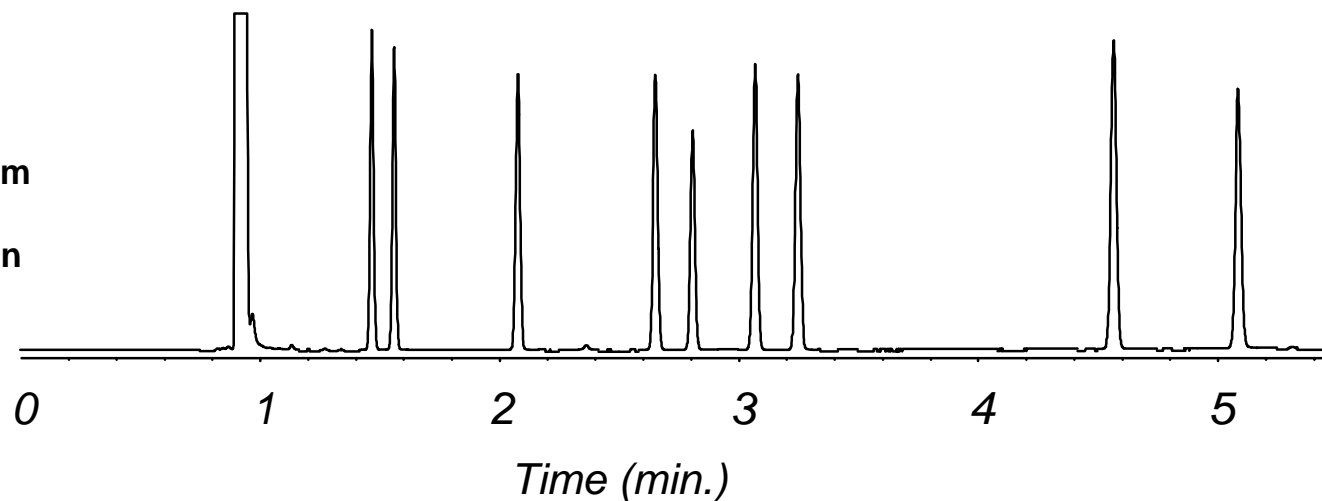


Start with the Right Phase

DB-1
15m x 0.32mm, 0.25 μ m
Oven:
40°C for 2 min
40-120°C at 5°C/min



DB-Wax
15m, 0.32mm, 0.25 μ m
Oven:
80-190°C at 20°C/min



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Change retention by changing temperature: K_c and Temperature

$$K_c = \frac{\text{conc. of solute in stationary phase}}{\text{conc. of solute in mobile phase}}$$

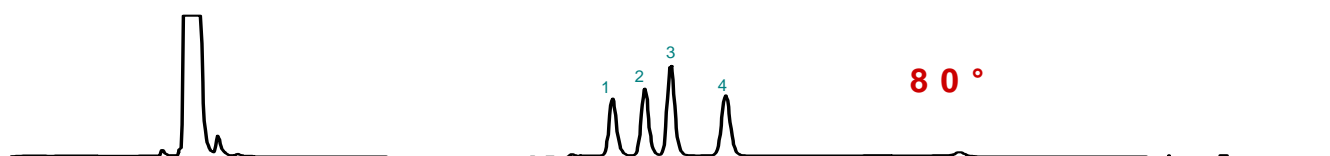
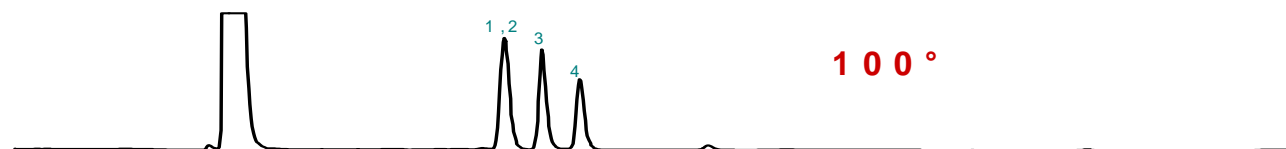
K_c decreases with an increase in temperature
Each solute's K_c may change at it's own rate



DB-WAX

Temperature Dependence

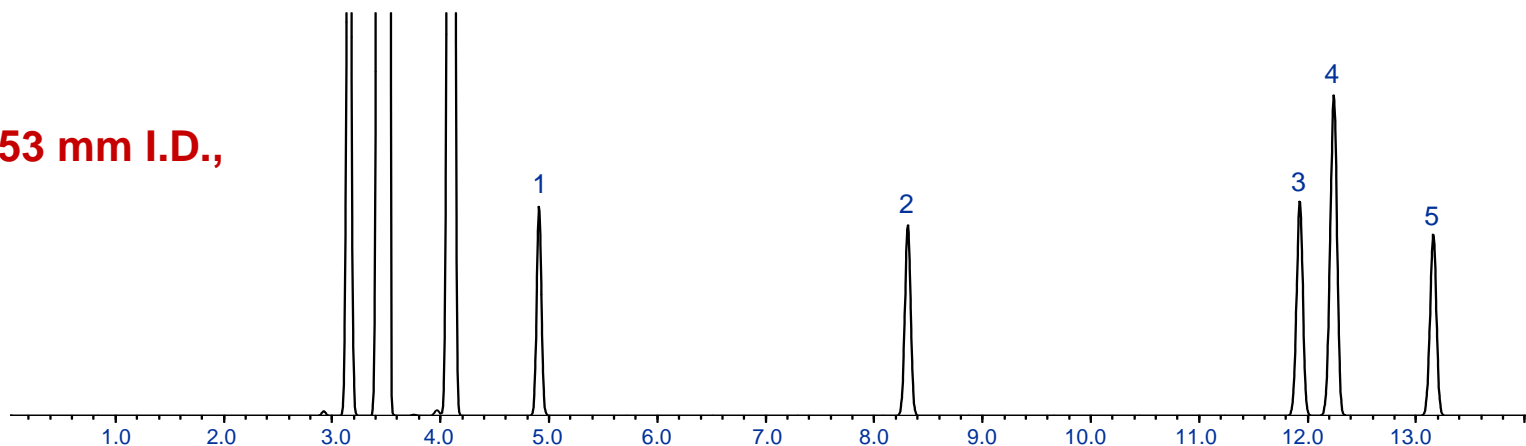
1. α -Terpinene
2. Dodecane
3. Limonene
4. 1,8-Cineole



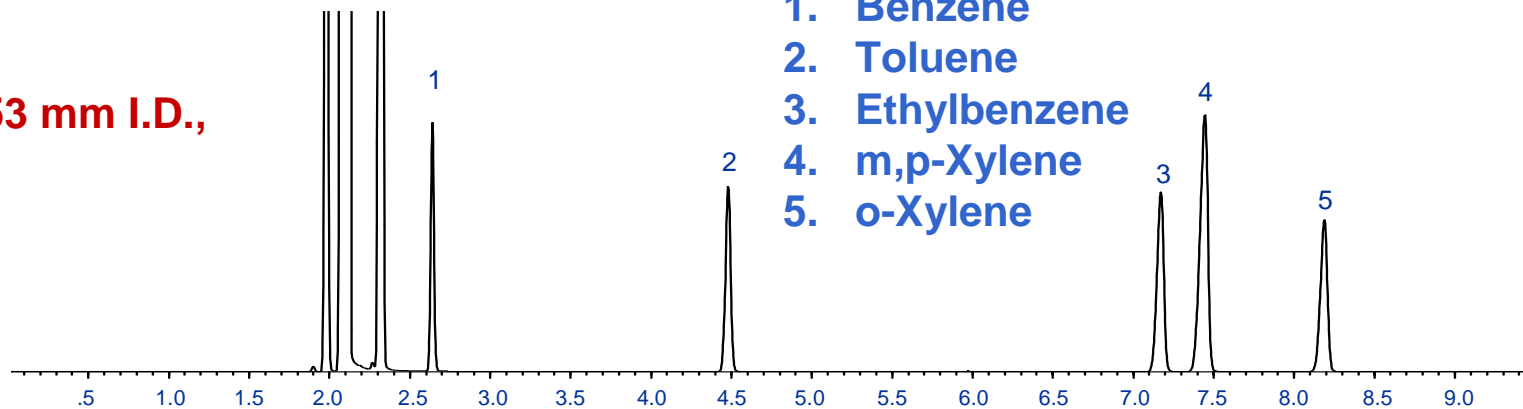
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Decreasing Film Thickness

DB-5
30 m, 0.53 mm I.D.,



DB-5
30 m, 0.53 mm I.D.,



1. Benzene
2. Toluene
3. Ethylbenzene
4. m,p-Xylene
5. o-Xylene

BTEX

Carrier: Helium, 36 cm/sec at 40°C

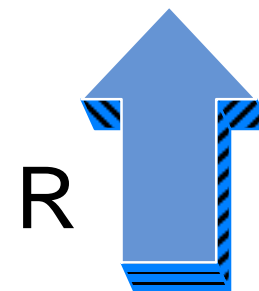
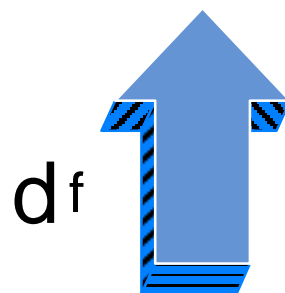
Oven : 40°C for 3 min, 5°/min to 100°C



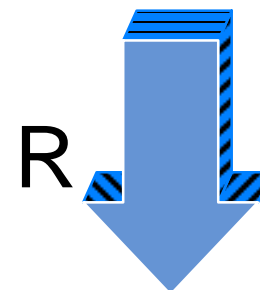
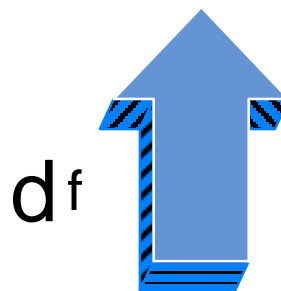
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Effect of Film Thickness on Resolution

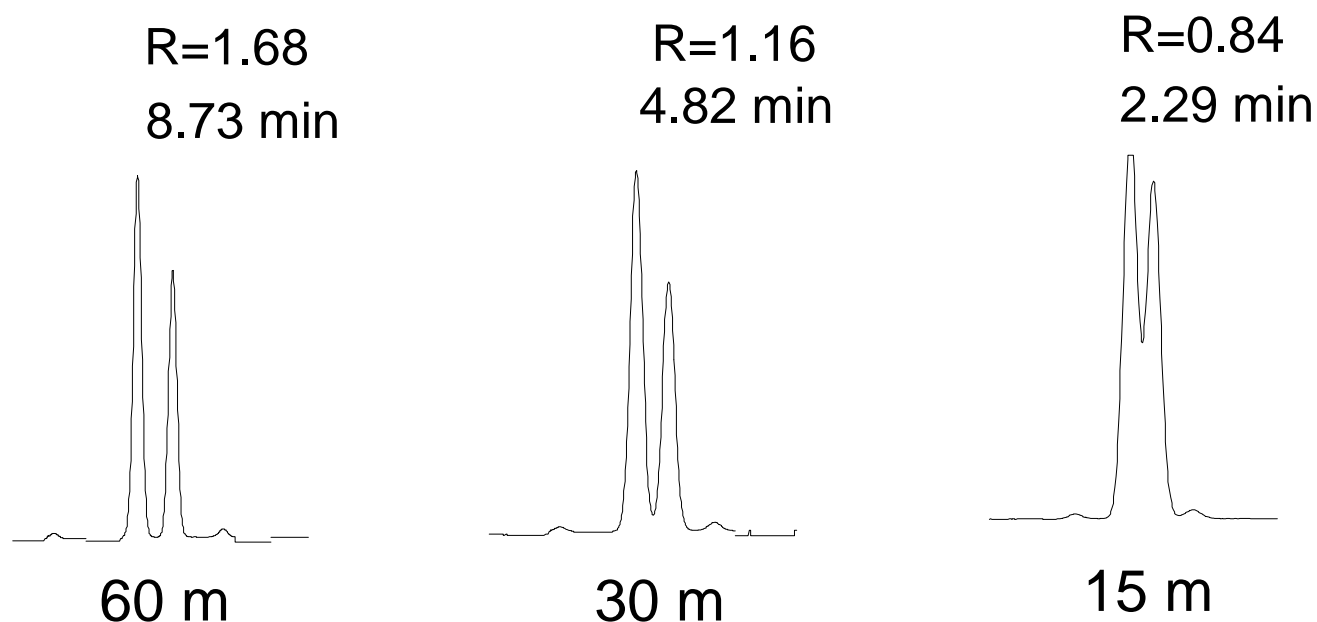
When solute $k < 5$



When solute $k > 5$



Column Length Resolution and Retention 210°C isothermal



Resolution is proportional to square root of length
Isothermal: Retention is proportional to length
Temperature program: 1/3-1/2 of isothermal values

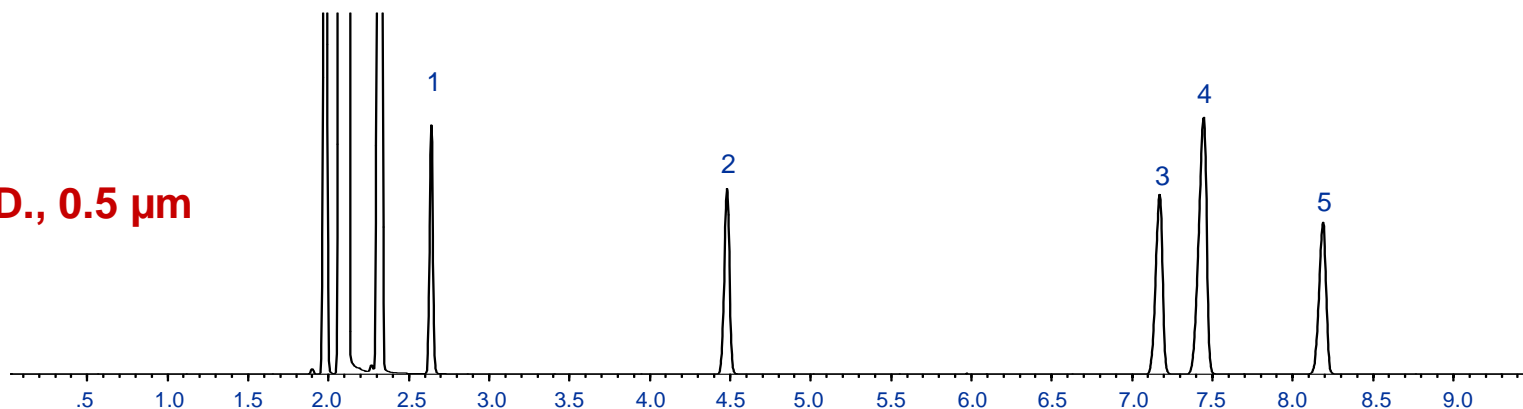


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Decreasing Column Length

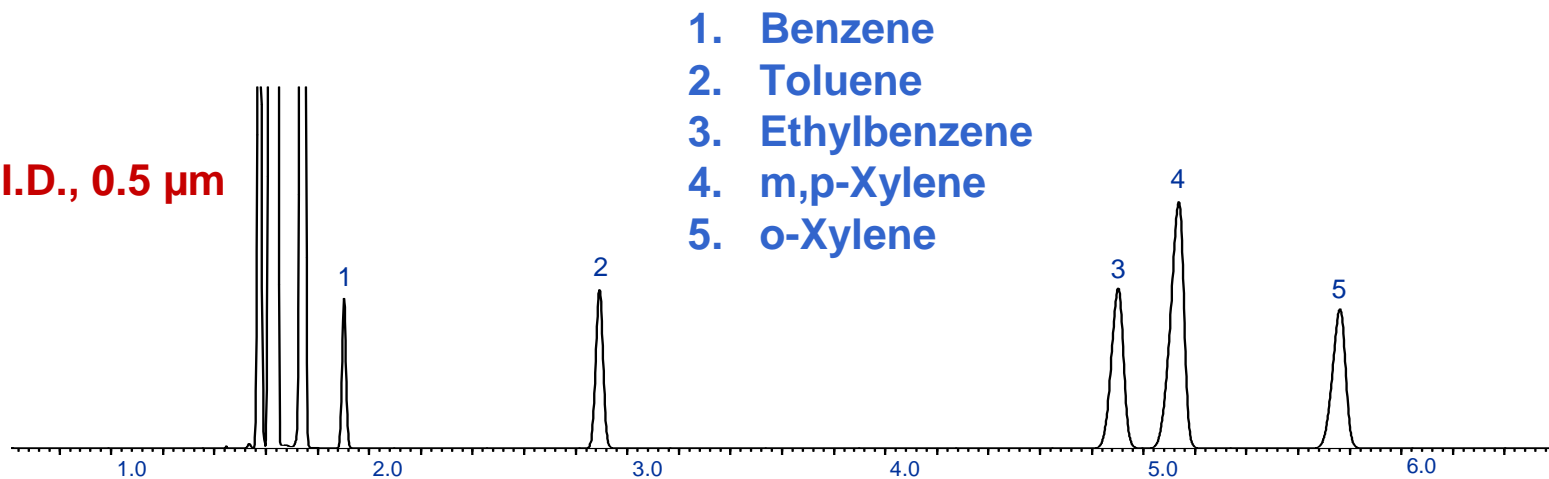
DB-5

0.53 mm I.D., 0.5 μ m



DB-5

0.53 mm I.D., 0.5 μ m



BTEX

Carrier: Helium, 36 cm/sec at 40°C

Oven : 40°C for 3 min, 5°/min to 100°C



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Column Diameter Theoretical Efficiency

	I.D. (mm)	N/m
k = 5	0.10	11905
	0.18	6666
	0.20	5941
	0.25	4762
	0.32	3717
	0.53	2242



Considerations For “Extreme” Dimension Changes

Shorten column length to decrease run time

Increase plates/meter by decreasing column diameter

For similar retention and selectivity keep the stationary phase and phase ratio ($\beta = r/d_f$) the same.



Considerations of using 0.1mm ID Columns

Carrier Gas

Temperature Program

Injection efficiency

Instrument considerations

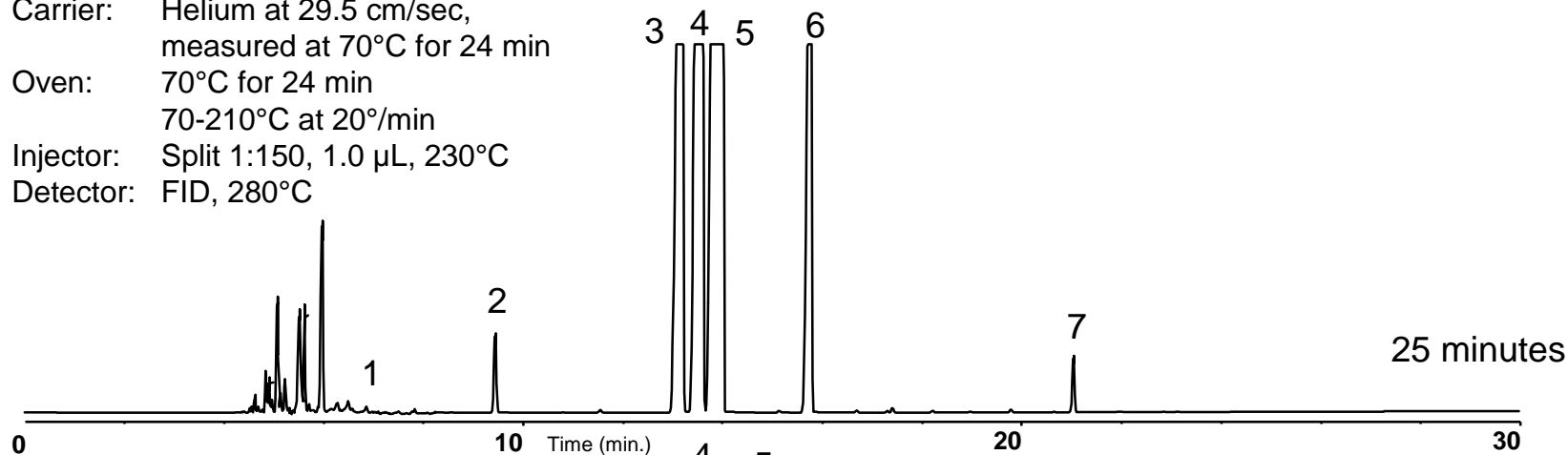
Working Range



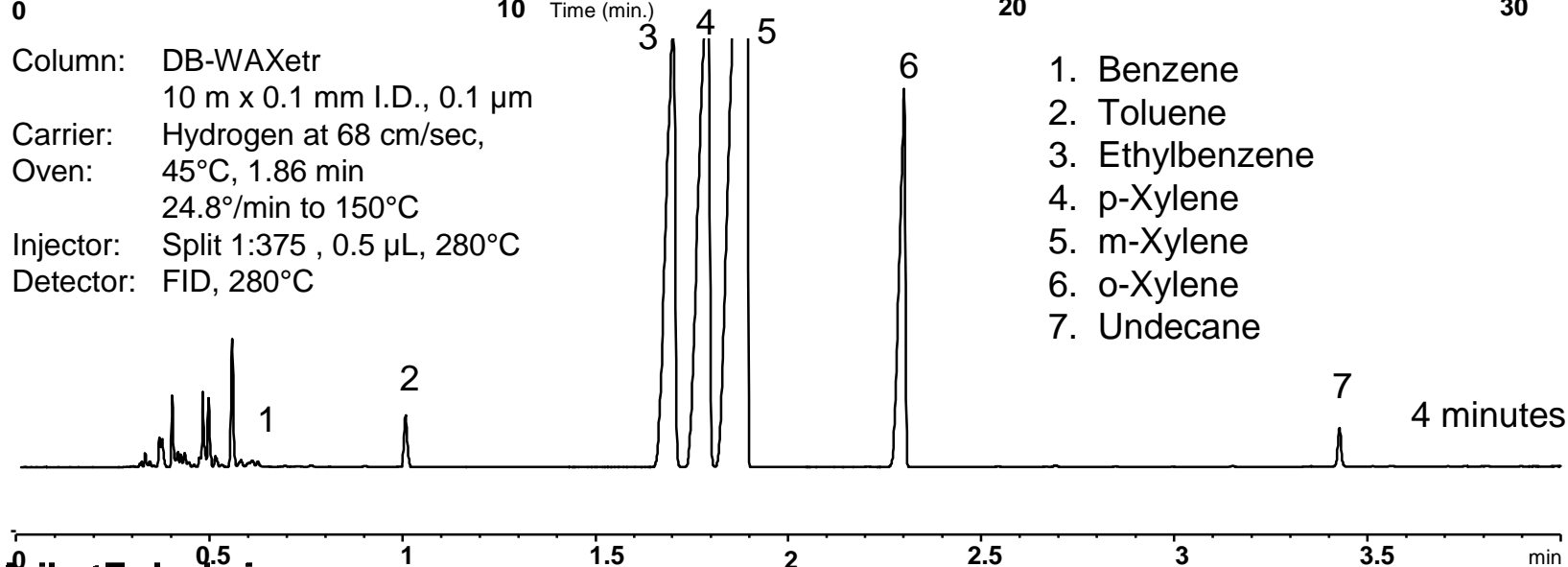
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Normal vs High Speed Analysis of Xylene (ASTM D2360)

Column: DB-WAXetr
60 m x 0.32 mm I.D., 0.25 μ m
Carrier: Helium at 29.5 cm/sec,
measured at 70°C for 24 min
Oven: 70°C for 24 min
70-210°C at 20°/min
Injector: Split 1:150, 1.0 μ L, 230°C
Detector: FID, 280°C



Column: DB-WAXetr
10 m x 0.1 mm I.D., 0.1 μ m
Carrier: Hydrogen at 68 cm/sec,
Oven: 45°C, 1.86 min
24.8°/min to 150°C
Injector: Split 1:375, 0.5 μ L, 280°C
Detector: FID, 280°C



- 1. Benzene
- 2. Toluene
- 3. Ethylbenzene
- 4. p-Xylene
- 5. m-Xylene
- 6. o-Xylene
- 7. Undecane



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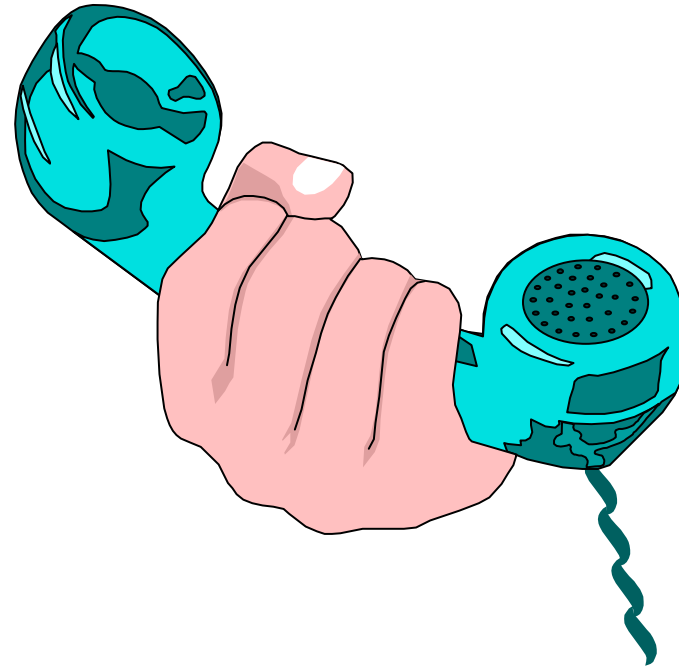
K_c and Temperature

K_c of analytes must be maintained
Temperature programs must be accurately
scaled to maintain relative analyte retention



Break Number 2

For Questions and Answers
Press *1 on Your Phone to
Ask a Question



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Method Translation Software

- What is Method Translation (MTL)?
 - A calculation technique used to scale a method to different column sizes, speeds, carrier gases, and detectors while maintaining the same elution order .
- How is Method Translation Done?
 - By using Agilent Technologies' free software for method translation (see note at end of this seminar section)
- Why does Method Translation Work?
 - The calculations maintain the same temperature ($^{\circ}\text{C}$) per void time in the oven program, which maintains elution order

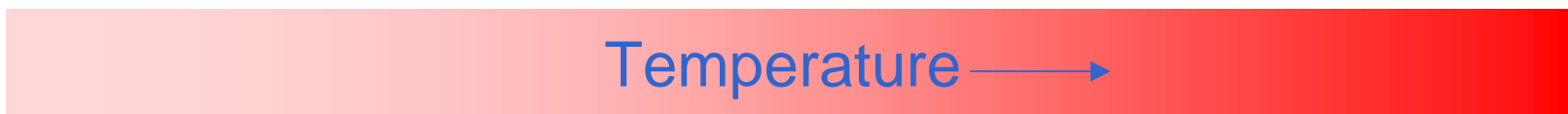


Temperature Programming

30m, 0.25mm ID



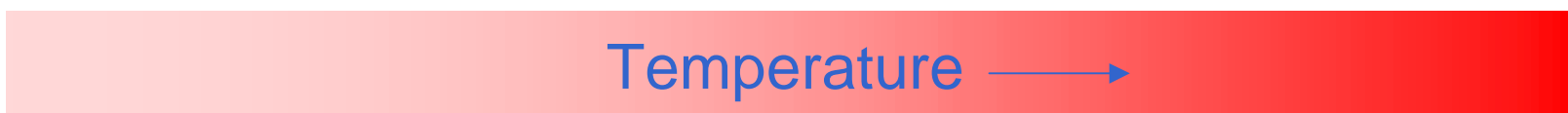
10m, 0.1mm ID



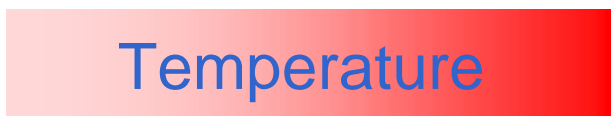
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Temperature Programming

30m, 0.25mm ID



10m, 0.1mm ID



Temperature program must be modified to give same temperature of elution (i.e. faster ramps, shorter hold times)



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Method Translation Software

Tool allowing GC methods to be translated to different conditions & maintain selectivity/resolution

new column configuration

different carrier gas

faster separation

Translates:

- inlet pressure, temp program, hold times

Benefits

- reduces methods development time
- help assess if GC method compatible with GC hardware



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Method Translation Software

GC Method Translation

Criterion: ☒ Translate Only ☐ Best Efficiency ☐ Fast Analysis ☐ None **Speed gain: 4.43545**

	Original Method	Translated Method																																				
Column																																						
Length, m	100.0	<input type="checkbox"/> 40.00																																				
Internal Diameter, μm	250.0	<input type="checkbox"/> 100.0																																				
Film																																						
Thickness, μm	0.500	<input type="radio"/> Unlock																																				
Phase Ratio	125.0	<input type="radio"/> 0.200																																				
		<input checked="" type="radio"/> 125.0																																				
Carrier Gas	Helium	<input type="checkbox"/> Hydrogen																																				
Enter one Setpoint																																						
Head Pressure, psi	40.000	85.242																																				
Flow Rate, mLn/min	1.7619	0.8810																																				
Outlet Velocity, cm/sec	62.33	194.77																																				
Average Velocity, cm/sec	23.77	42.17																																				
Hold-up Time, min	7.01259	1.58103																																				
Outlet Pressure (absolute), psi	14.696	<input type="checkbox"/> 14.696																																				
Ambient Pressure (absolute), psi	14.696	<input type="checkbox"/> 14.696																																				
Oven Temperature 3-ramp Program	<table border="1"> <thead> <tr> <th>Ramp Rate</th> <th>Final Temp.</th> <th>Final Time</th> </tr> <tr> <th>$^{\circ}\text{C}/\text{min}$</th> <th>$^{\circ}\text{C}$</th> <th>min</th> </tr> </thead> <tbody> <tr> <td>Initial</td> <td>35.00</td> <td>13.000</td> </tr> <tr> <td>Ramp 1</td> <td>10.000</td> <td>45.00</td> </tr> <tr> <td>Ramp 2</td> <td>1.000</td> <td>60.00</td> </tr> <tr> <td>Ramp 3</td> <td>2.000</td> <td>220.00</td> </tr> </tbody> </table>	Ramp Rate	Final Temp.	Final Time	$^{\circ}\text{C}/\text{min}$	$^{\circ}\text{C}$	min	Initial	35.00	13.000	Ramp 1	10.000	45.00	Ramp 2	1.000	60.00	Ramp 3	2.000	220.00	<table border="1"> <thead> <tr> <th>Ramp Rate</th> <th>Final Temp.</th> <th>Final Time</th> </tr> <tr> <th>$^{\circ}\text{C}/\text{min}$</th> <th>$^{\circ}\text{C}$</th> <th>min</th> </tr> </thead> <tbody> <tr> <td>Initial</td> <td>35.00</td> <td>2.931</td> </tr> <tr> <td>Ramp 1</td> <td>44.354</td> <td>45.00</td> </tr> <tr> <td>Ramp 2</td> <td>4.435</td> <td>60.00</td> </tr> <tr> <td>Ramp 3</td> <td>8.871</td> <td>220.00</td> </tr> </tbody> </table>	Ramp Rate	Final Temp.	Final Time	$^{\circ}\text{C}/\text{min}$	$^{\circ}\text{C}$	min	Initial	35.00	2.931	Ramp 1	44.354	45.00	Ramp 2	4.435	60.00	Ramp 3	8.871	220.00
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Sample Information None																																						



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Injector Efficiency

Narrow columns generate narrow peaks

Injection band must be narrow to take advantage of the column efficiency

This usually means using split injection with high split ratios



Instrument Considerations

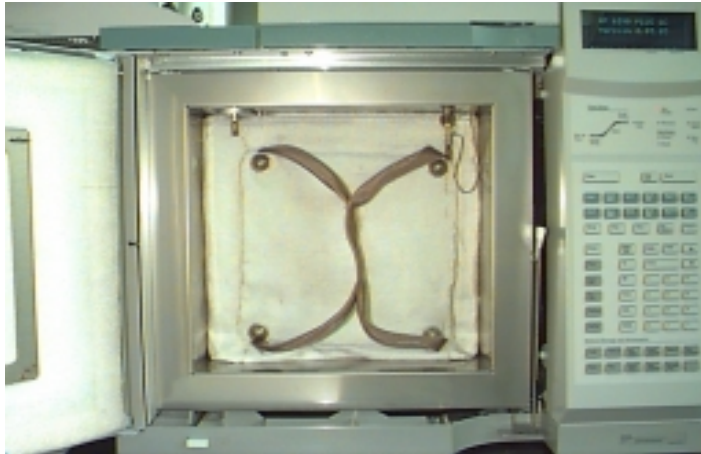
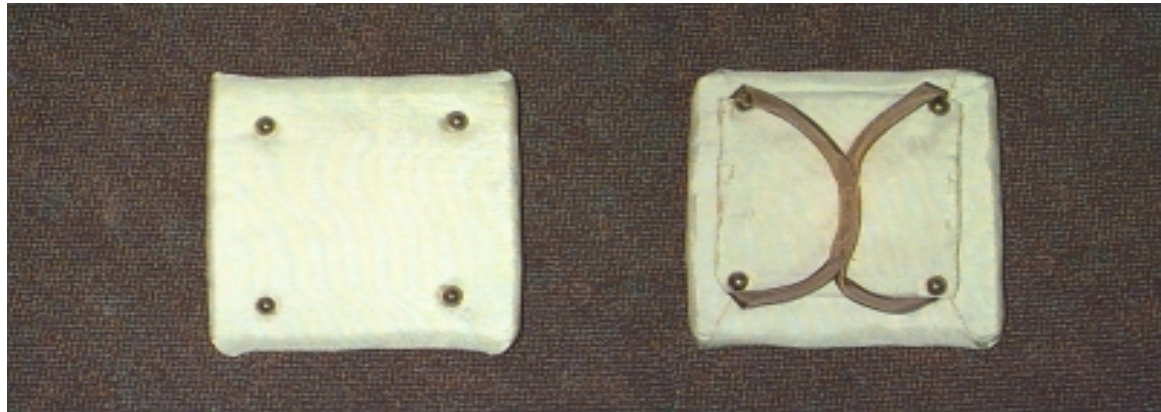
6890/6850 Optimized For Fast GC

- Sample Introduction/ Inlets
 - fast automatic injector (0.1 sec injection time)
 - high pressure, high flow capability
 - 100 psi inlet (optional 150 psi inlet for 6890)
 - up to 1000 mL/min flow rates
- Fast Oven - up to 120 °C/min ramp rates
- Fast detector electronics
 - FID, NPD, FPD to 200 Hz
 - ECD to 50 Hz



Instrument Considerations

Oven insert for the 6890



- The insert reduces the 6890 effective oven volume by 50%.
- Allows 120V 6890 to achieve ramp rate equal to a 240V 6890 or a 6850.
- Allows faster oven cool down over std 6890.
- Part No. G2646-60500



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6890/6850 Oven Ramp Rates

		“Standard”			“Fast”			“Turbo”
Temp. Range (°C)		6890 120V	6890 240V	6890 120V Insert	6850 120V			6890 240V Insert
50 to 70		75	120	120	120			120
70 to 115		45	95	95	95			120
115 to 175		40	65	65	65			110
175 to 300		30	45	45	45			80
300 to 450		20	35	35	35*			65

* Maximum temperature for 6850 is 350 °C

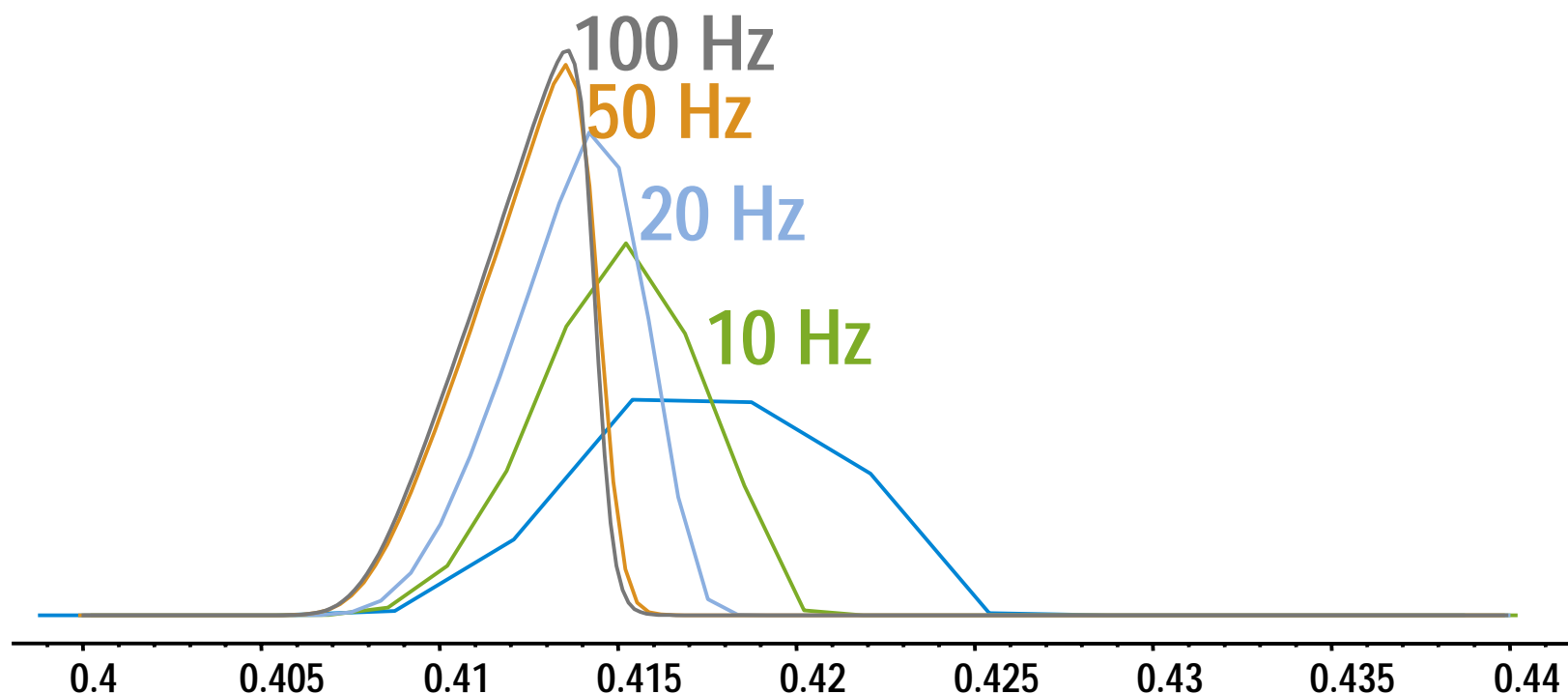
- Allows sharing of fast GC methods across platforms



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Effect of Data Rate on Peak Height

Set data rate to give 10 points across halfwidth of peak. Peak is 0.23 sec at 1/2 width, so use 50 Hz



Working Range

$$W = Q_s \div Q_o$$

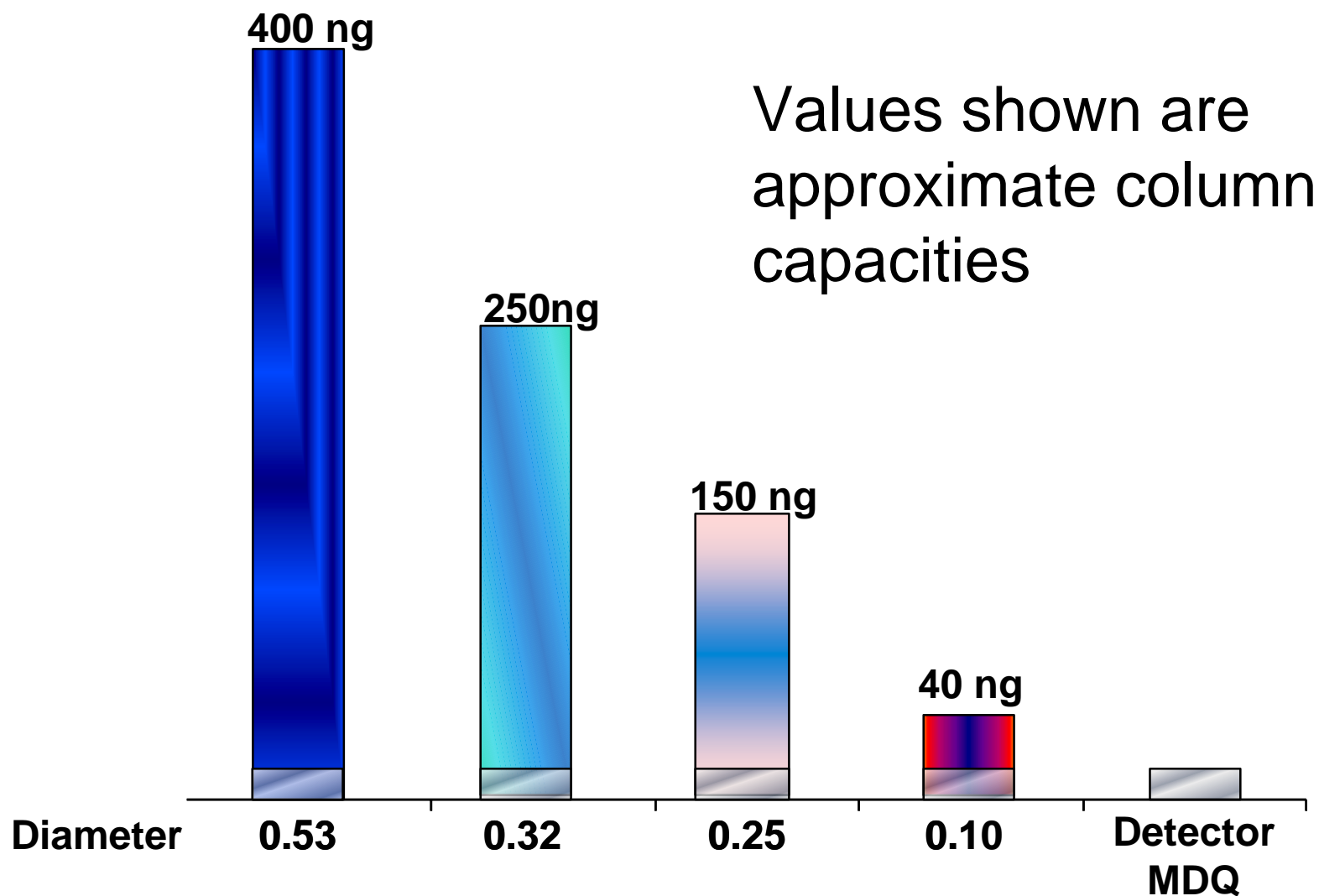
Q_s = maximum column capacity

Q_o = minimum amount that can be reliably detected

Column capacity is proportional to column diameter
Column diameter will have little effect on detector sensitivity



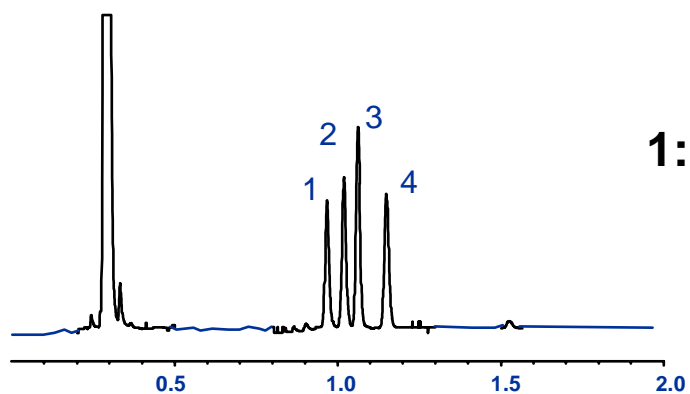
Working Range



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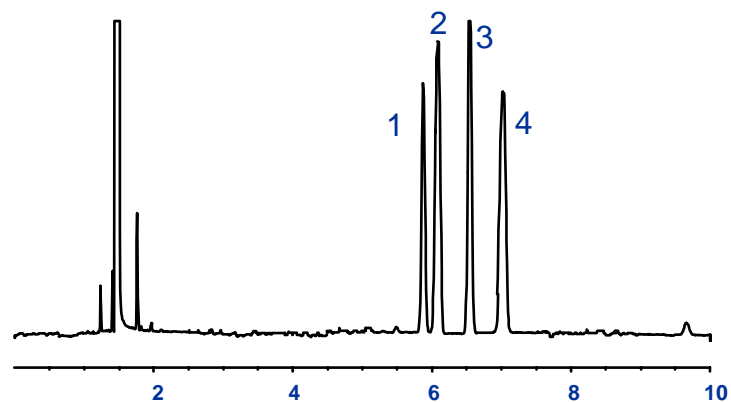
Capacity: Effect on Resolution

10 m x 0.1 mm, 0.2 μ m

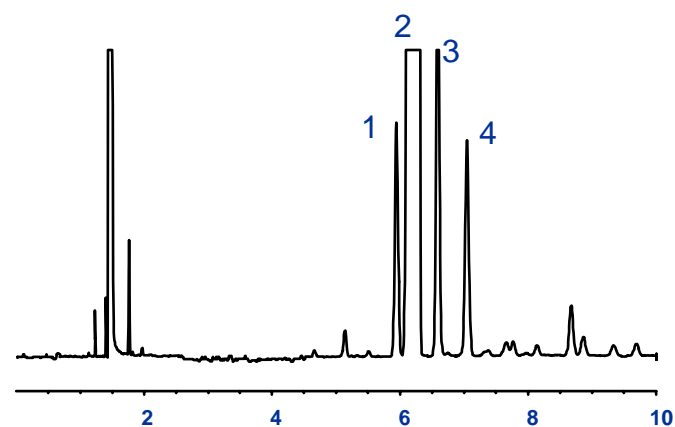
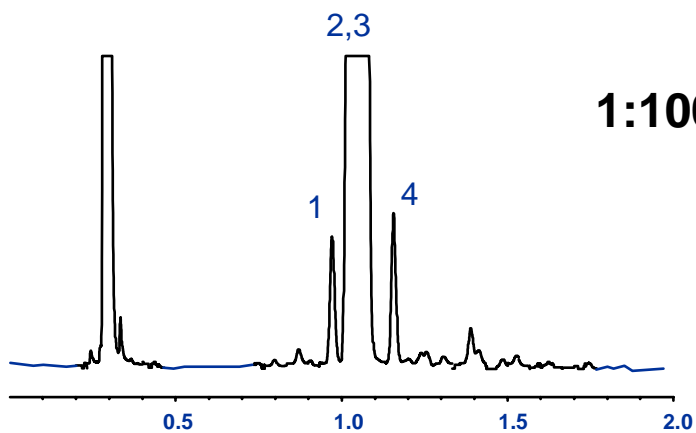


1:1:1:1

30 m x 0.25 mm, 0.5 μ m



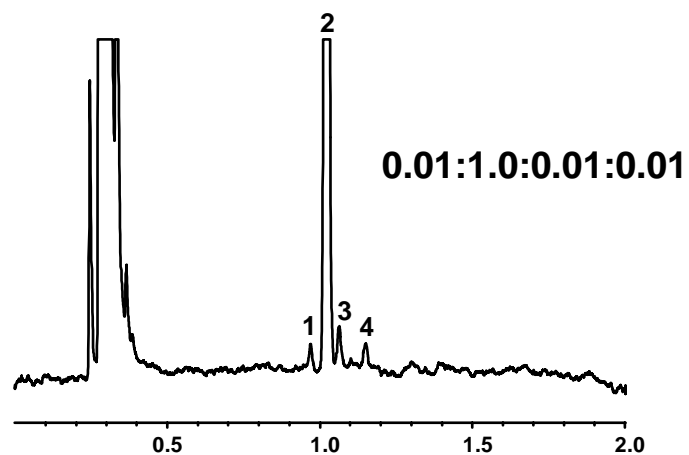
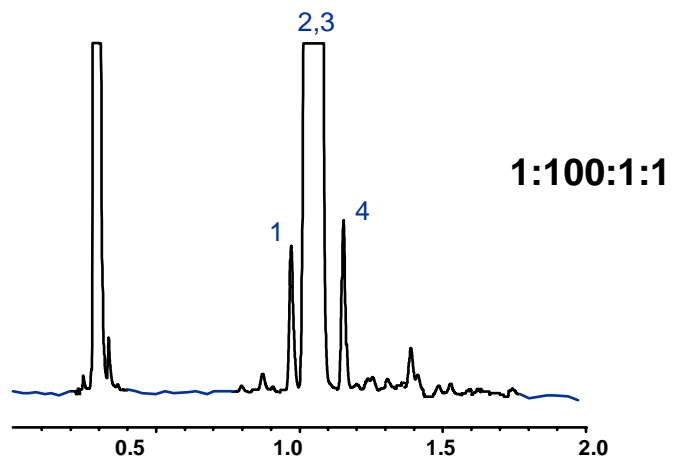
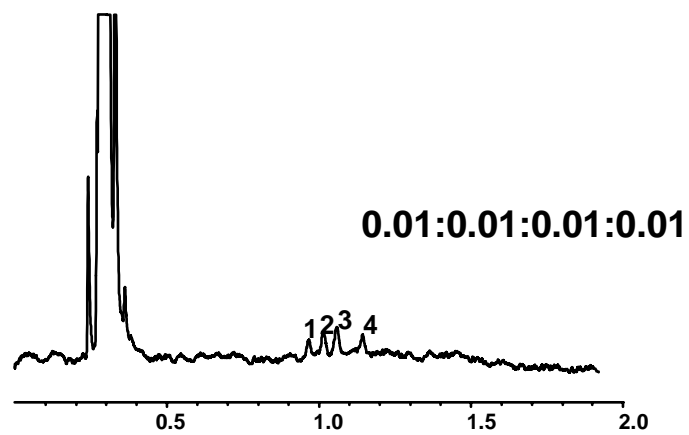
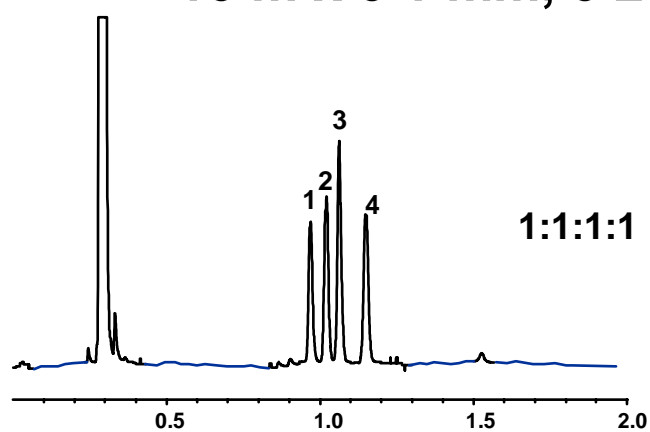
1:100:1:1



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Is Dilution the Solution?

10 m x 0.1 mm, 0.2 μ m



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Requirements for Microbore High Speed GC Analysis

- Rapid injection speed (100 ms) - split
- High pressure inlet and pressure regulators (150 psi max)
- H₂ carrier gas (for efficiency, speed and practicality)
- Fast oven programming rate
- Fast detector sample rate (200 Hz)



Microbore High Speed GC Analysis Considerations

- Column capacity (25 - 40 ng)
- Carrier gas - type, pressure and plumbing
- GC System
- Method translation
- Discrimination
- Sensitivity



Retention Time Locking



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What does RTL provide?

Compound retention times become...


Permanent and Universal

Unchanging RTs

- Fast & Easy Compound Identification
- Better QA/QC
- Simpler SOPs
- Proprietary Locked Compound Databases

Shared Methods

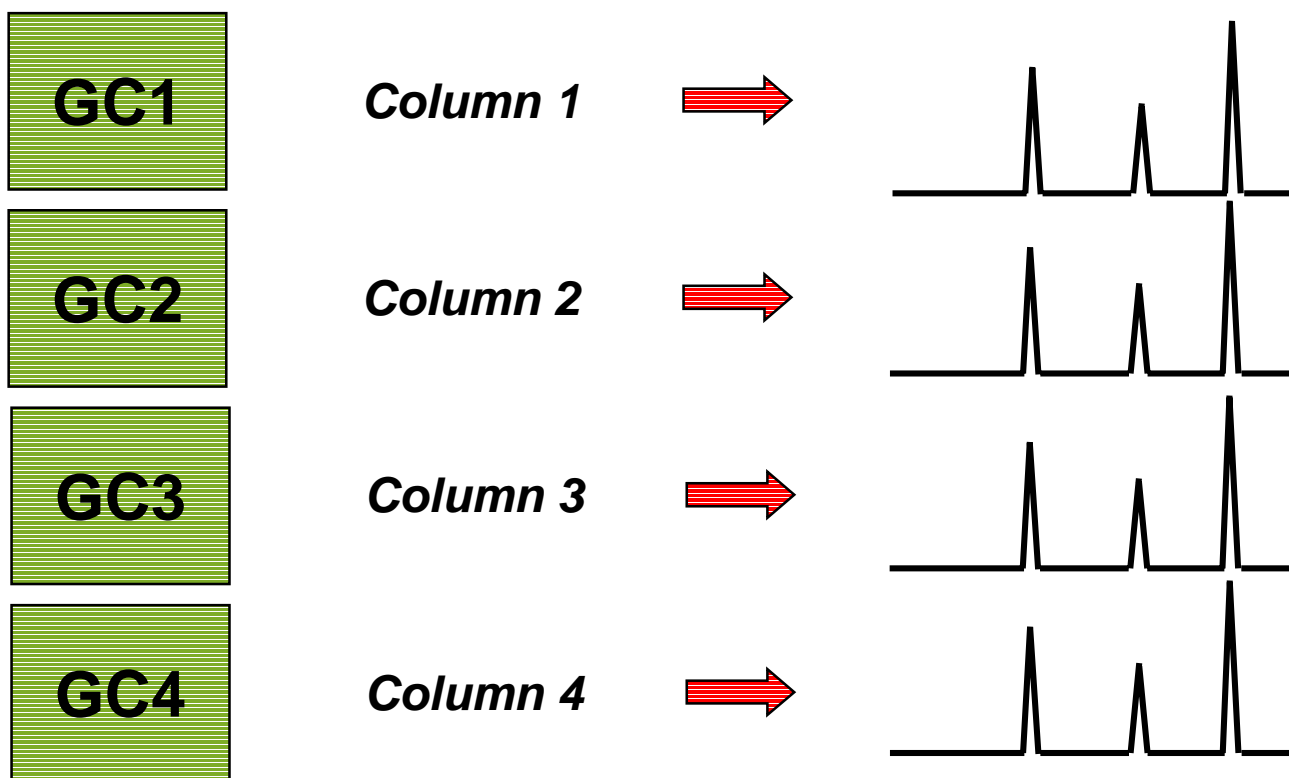
- Rapid and Simple Method exchange
- Intra- & Inter- laboratory SOPs
- Higher Assuredness / Confidence
- Shared Locked Compound Databases



Lowered costs, higher quality and increased productivity

Retention Time Locking

Retention times match from column-to-column and instrument-to-instrument to 0.030 min or better...



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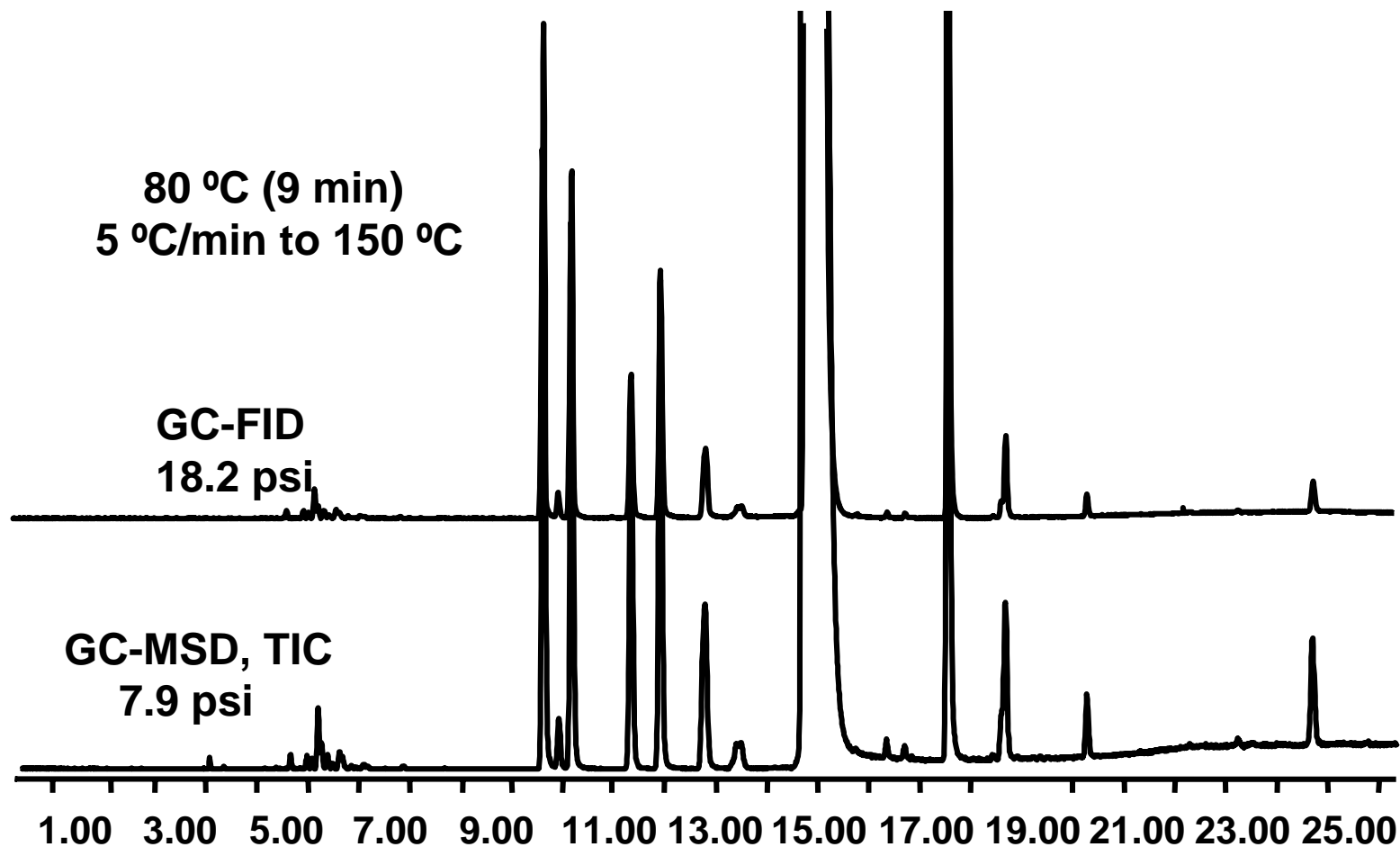
Retention Time Locking

- Can be used with the Method Translation Software
 - different column dimensions (0.53, 0.32, 0.18, 0.10, etc.)
 - different detectors (MSD, AED, FID, etc.)
- Higher level of confidence in peak identification, especially between different instruments and different laboratories.
- Reduction in data analysis time.
- Greatly reduces problems associated with adjusting RT windows in method, peak misidentification, timed events such as valve switching, adjusting “special” integration events for problem areas in chromatogram.



RT-Locked GC-FID and GC-MSD

Impurities in styrene analysis



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Same Method, Different Column Sizes

- **Why?**
 - **530 μm - Ease of use, robustness**
 - **320 μm - Good all around compromise**
 - **250 μm - Meets MSD flow requirements**
 - **100 μm - Fast GC with no resolution loss**
- **How**
 - **Use MTL software to calculate changes in pressure**
 - **Generate new RTL calibration for new column**
 - **Lock system**



Designing Global Methods...

- Designed for use in multiple:
 - instruments
 - configurations (FID, AED, micro-ECD, MSD, PTV, S/S)
 - locations
- Provides known retention times and elution order
 - Easy setup and maintenance of method
 - Easy data analysis and interpretation
- Accurately scalable
 - Adjustable for optimal speed/resolution/capacity



With Global Methods...

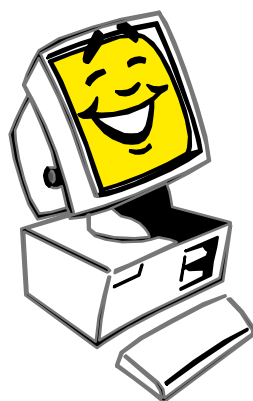
- Faster method transfer & validation
- Reduce / eliminates need to update RTs in calibration tables
- Easier review of data
- Faster identification of unknowns
- More efficient troubleshooting
- Scalable methods link chromatography in QA labs with that in support labs
- Scalable methods allow optimization of speed, resolution and capacity of method for specific analysis



Agilent J&W Technical Support

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