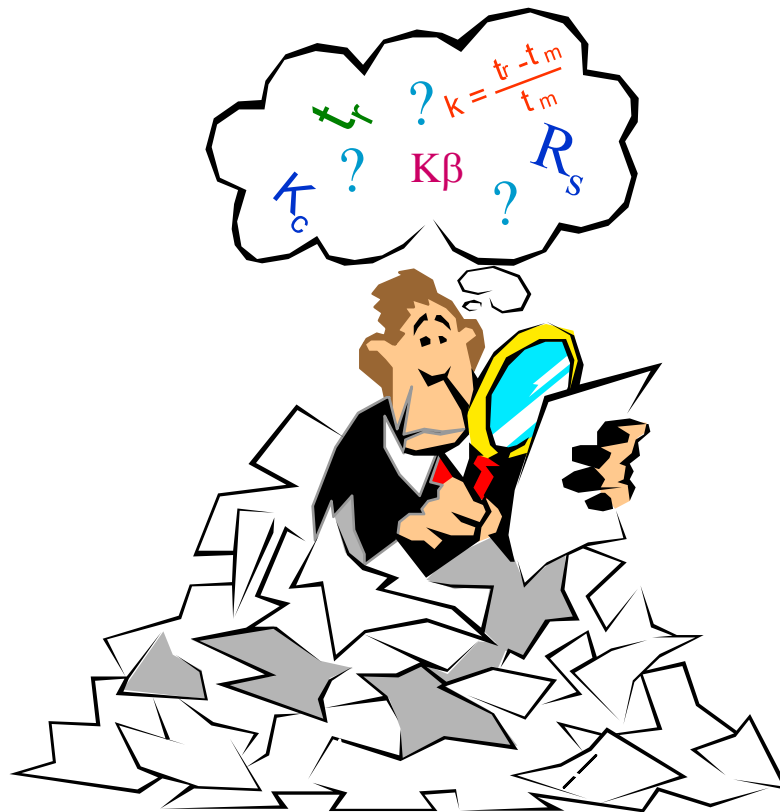
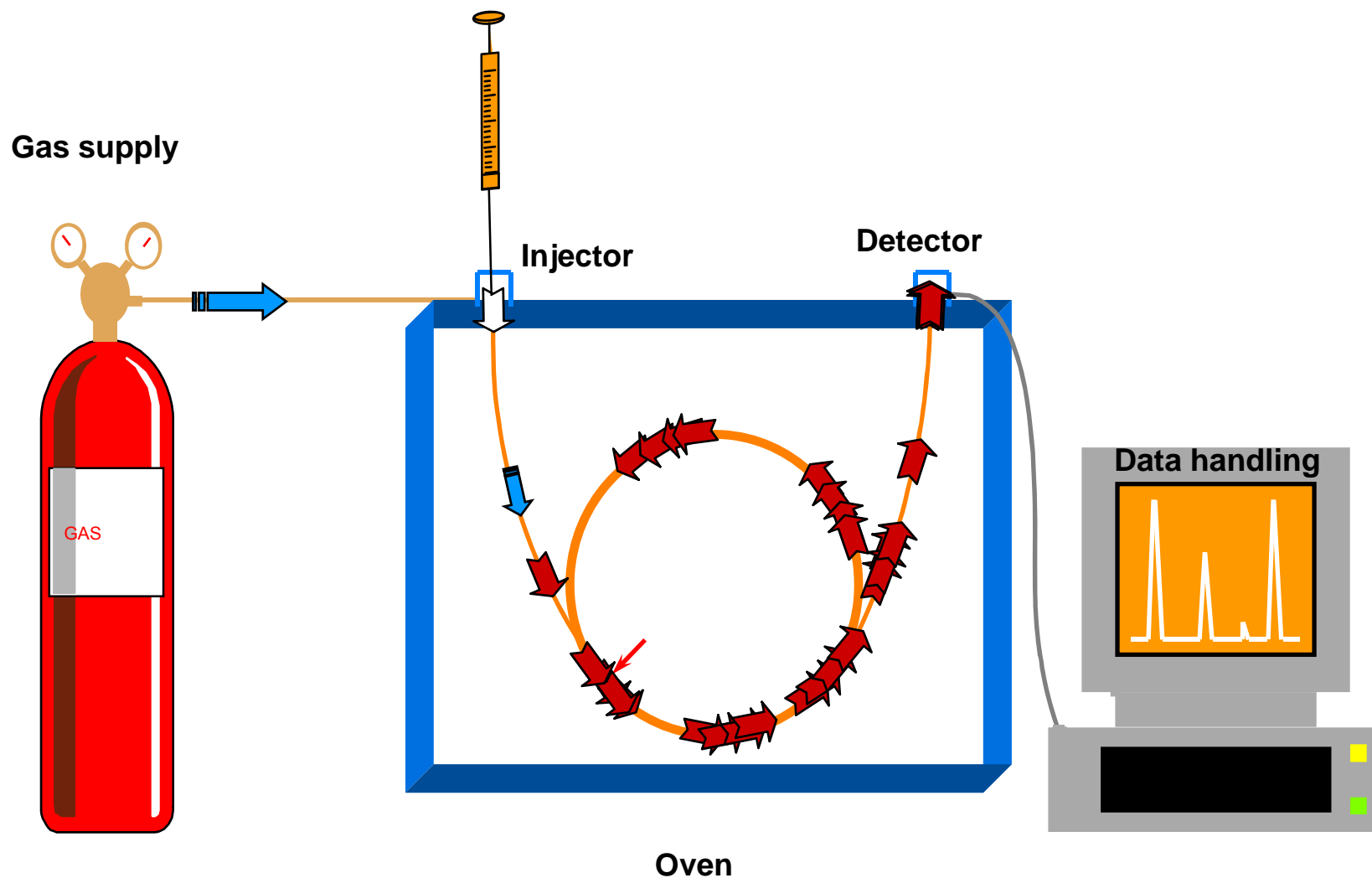


# Introduction to Capillary GC



# Typical GC System



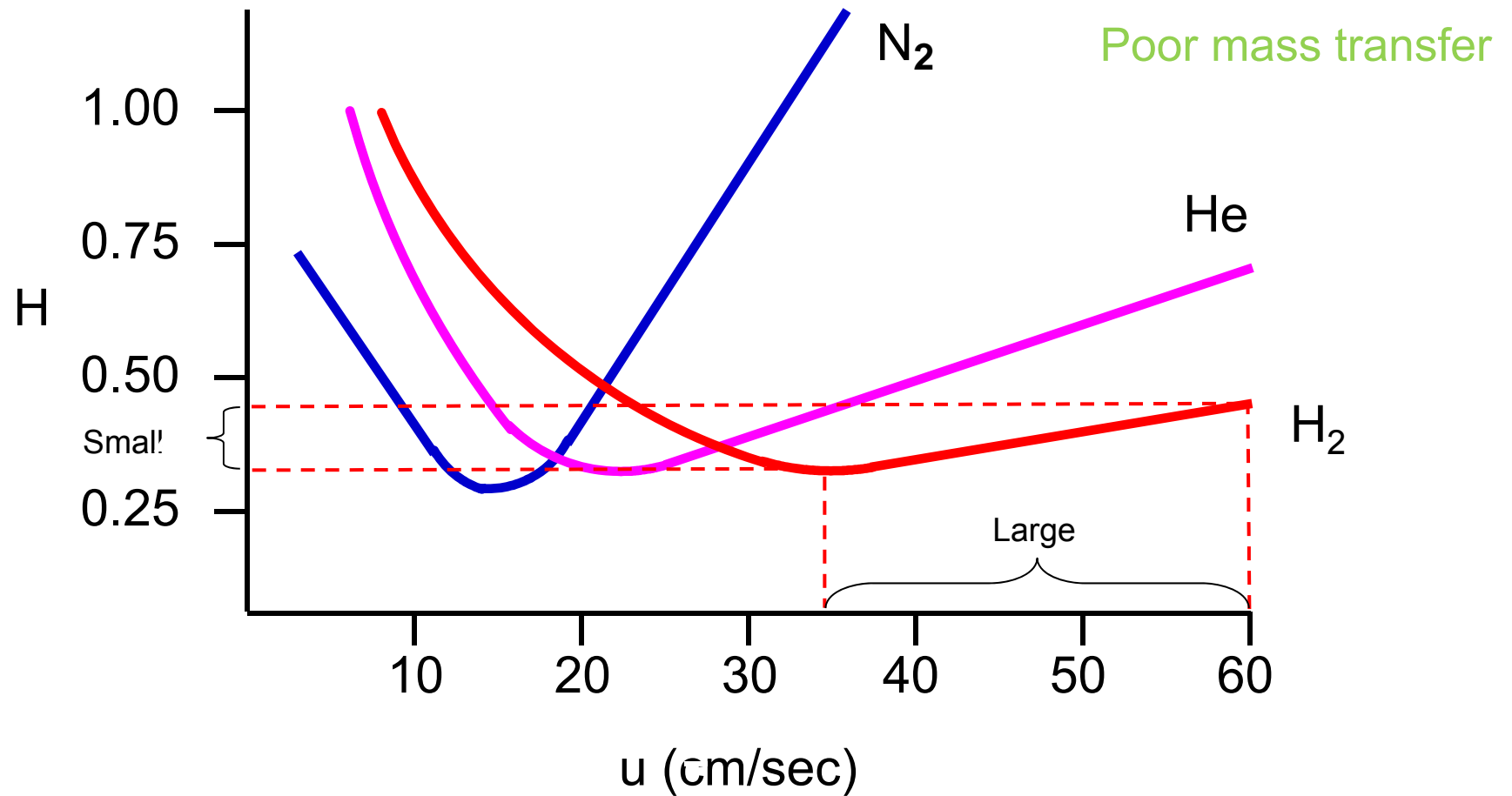
# CARRIER GAS

Carries the solutes down the column

Selection and velocity influences efficiency and retention time

# VAN DEEMTER CURVES

Excess diffusion



# CARRIER GAS

Type	Velocity Range ( $u_{\text{opt}}$ – OPGV)
Nitrogen	8-16
Helium	20-40
Hydrogen	30-55

$\mu_{\text{opt}}$  Optimal Carrier Gas Velocity

OPGV Optimal Practical Gas Velocity

1.5-2 times the  $\mu_{\text{opt}}$

# SAMPLE INJECTION

## Goals:

Introduce sample into the column

Reproducible

No efficiency losses

Representative of sample

# Sample Introduction

Purpose: To introduce a representative portion of sample onto the column in a reproducible manner, while minimizing sample bandwidth

## Syringe Injection

## Autosampler injection

## Valve Injection

- Gas sampling valve
- Liquid sampling valves



Objective: The sample must not be chemically altered , unless desired (e.g., derivatization). Success is no contamination, degradation, or discrimination.

# Types of Inlets

Purged Packed

Split / Splitless

Cool On Column

Programmable Temperature Vaporization

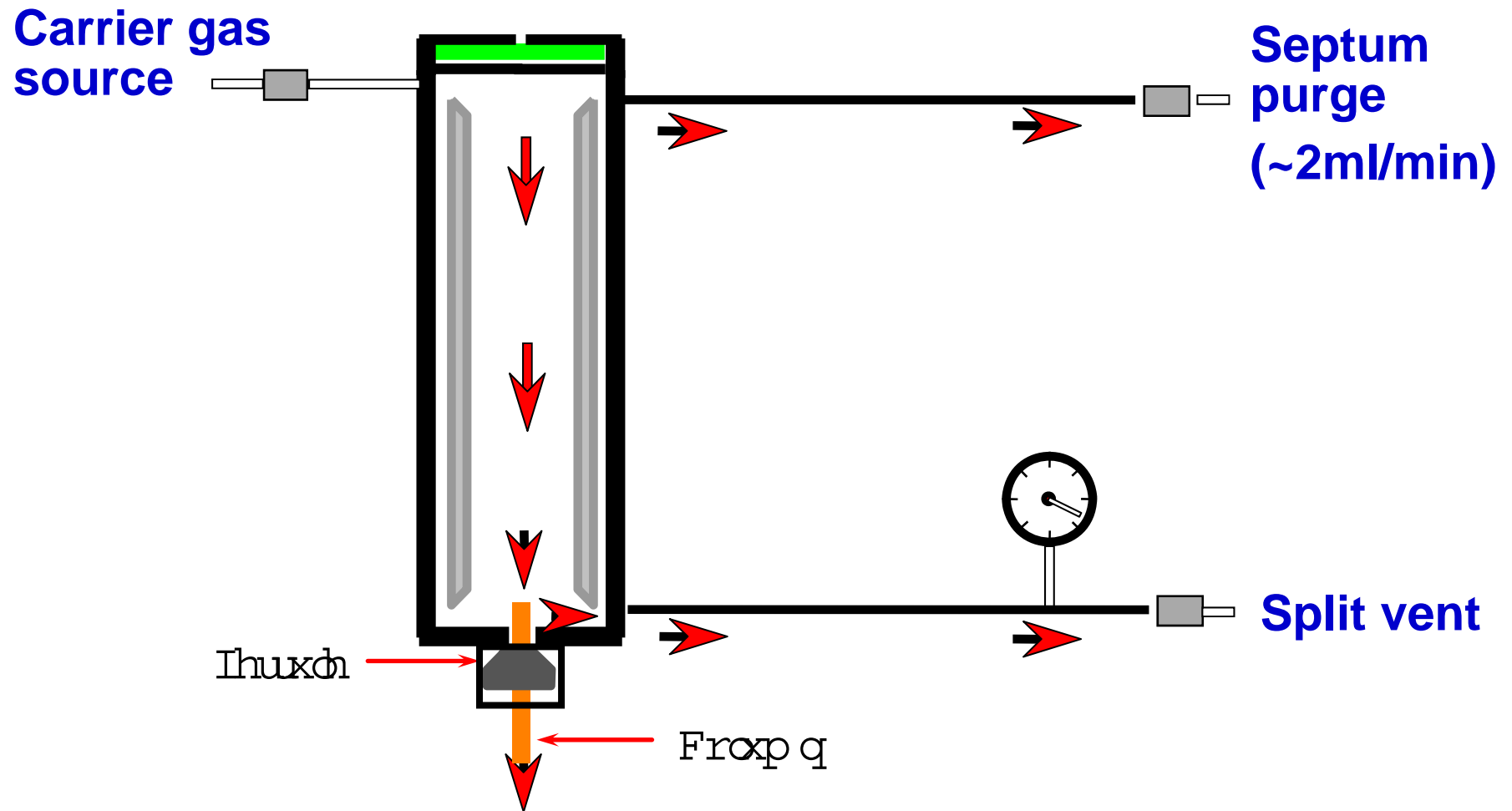
Volatiles Interface

Multi Mode Inlet





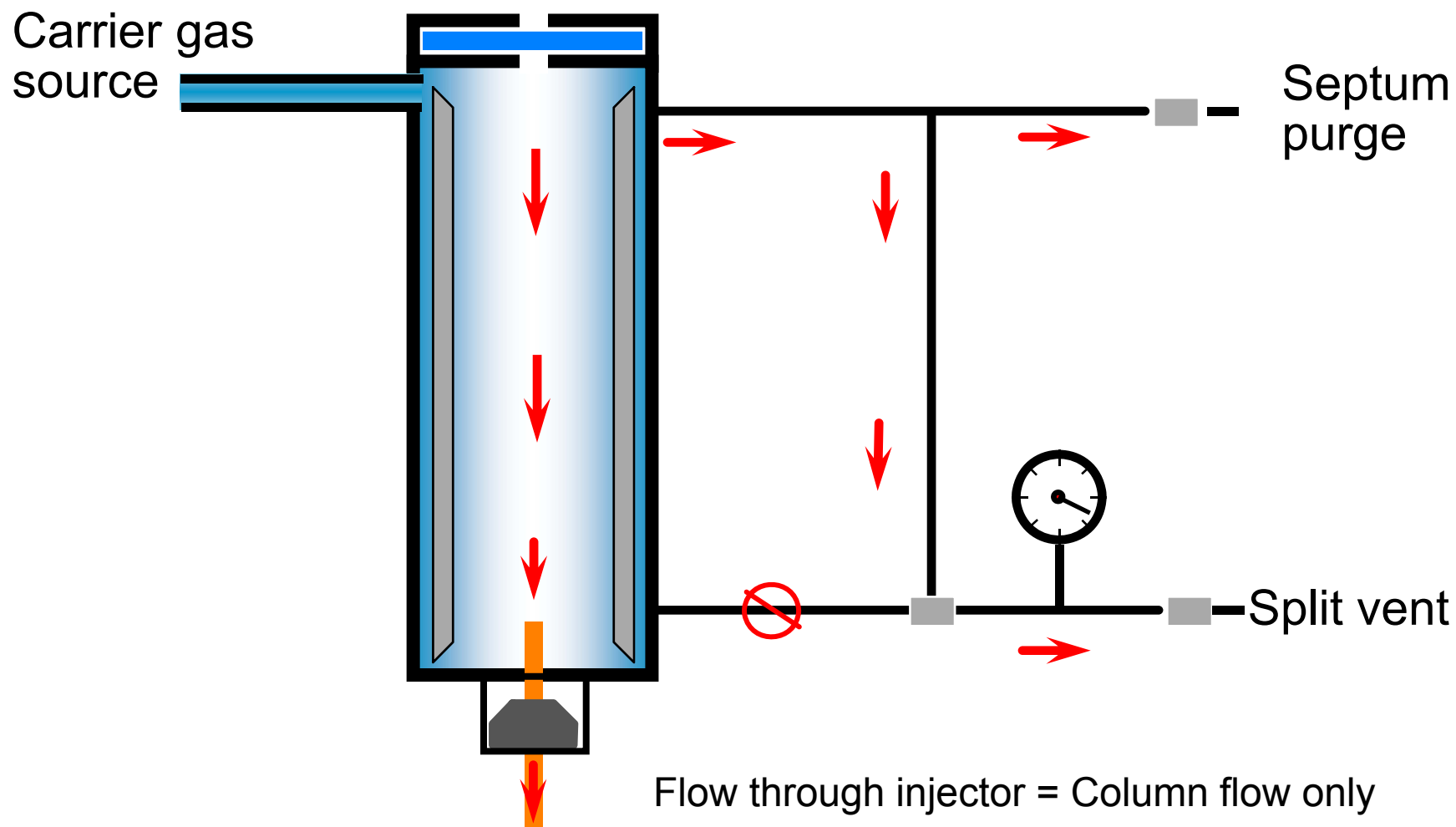
# SPLIT/SPLITLESS INJECTOR



**Flow through injector = Column flow + Split Vent Flow**

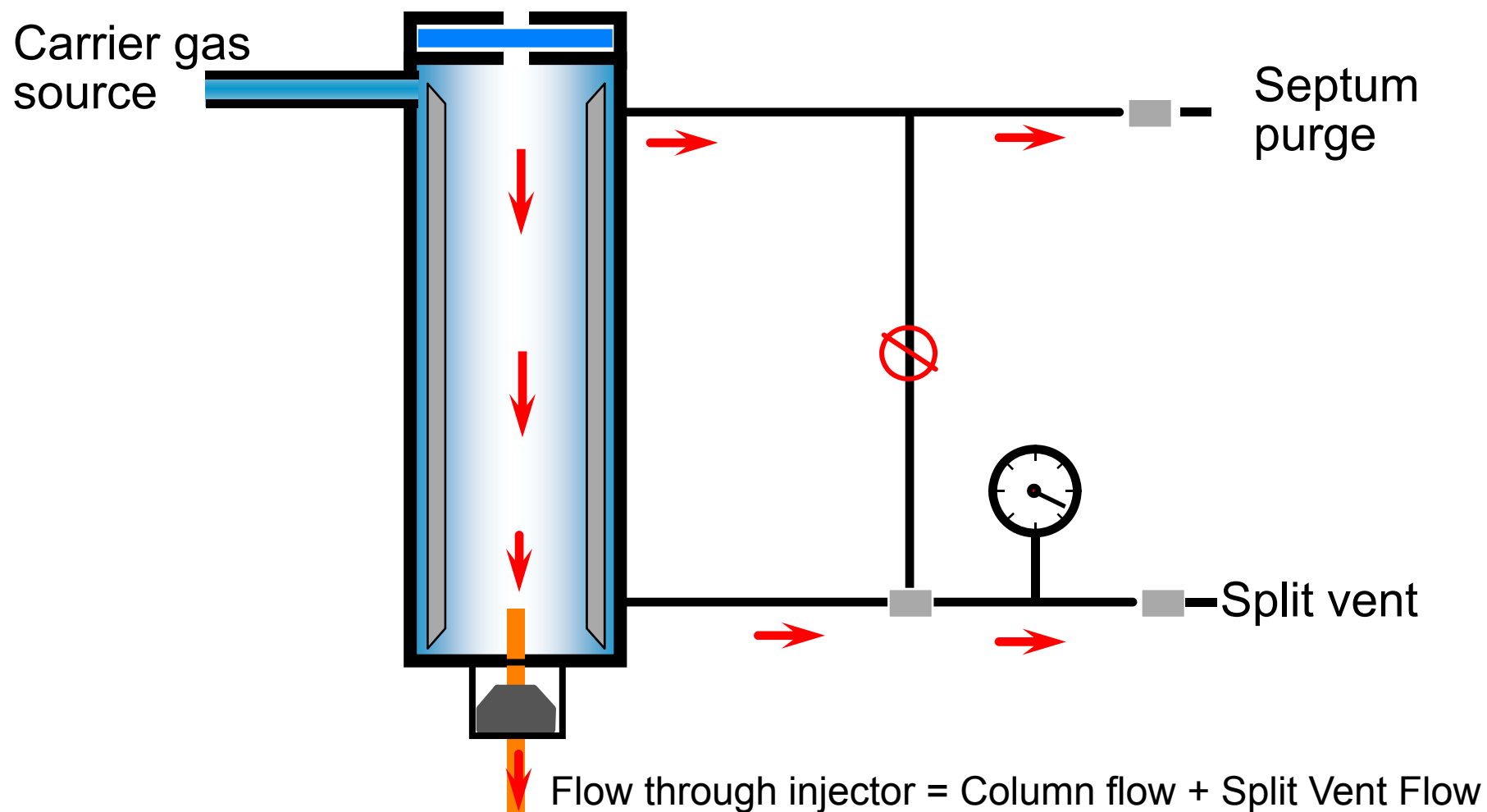
# Splitless Injector

## Purge Off At Injection



# Splitless Injector

## Purge On After Injection



# DETECTORS

## Purpose:

Responds to some property of the solutes

Converts the interaction into a signal

Immediate

Predictable



# Detectors

<b>Detector</b>	<b>Dynamic Range</b>		<b>MDL</b>
TCD	$10^5$	Universal	400 pg Tridecane
FID	$10^7$	Responds to C-H bonds	1.8 pg Tridecane
ECD	$5 \times 10^5$	Responds to free electrons	6 fg/mL Lindane
NPD	$10^5$	Specific to N or P	0.4 pgN/s 0.06 pg P /s
FPD	$10^3$ S, $10^4$ P	Specific to S or P	60 fg P/s 3.6 pg S/s
SCD	$10^4$	Specific & Selective to S	0.5 pg S/s
NCD	$10^4$	Specific & Selective to N	3 pg N/s
MSD		Universal	S/N 400:1 1 pg/uL OFN

# DATA HANDLING

## Converts the detector signal into a chromatogram

- Integrator
- Software Program

# COMPOUND REQUIREMENTS FOR GC

Only 10-20% of all compounds are suitable for GC analysis

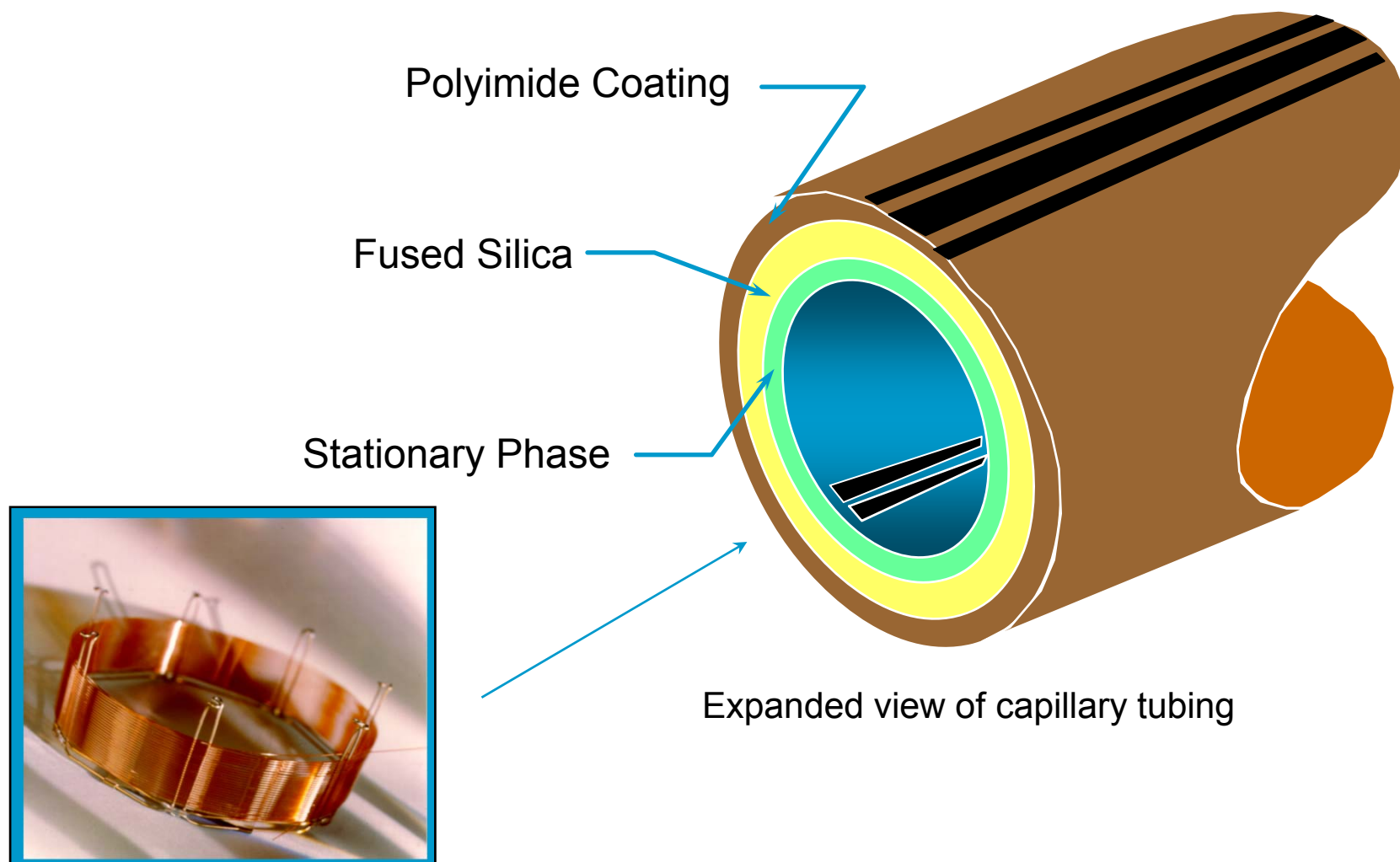
The compounds must have:

- ✓ Sufficient volatility
- ✓ Thermal stability

**NO** Inorganic Acids and Bases

Be mindful of salts!

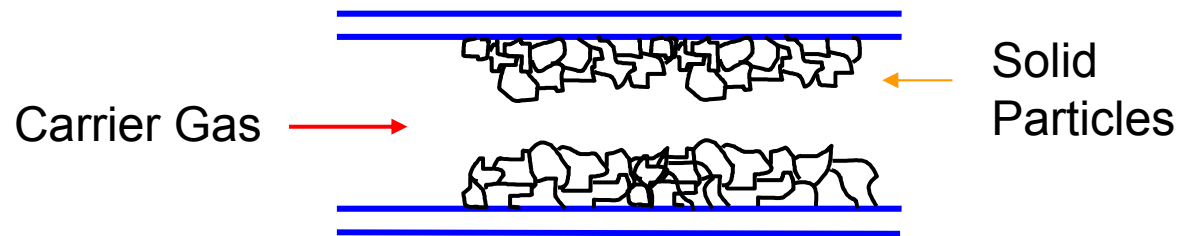
# Typical Capillary Column



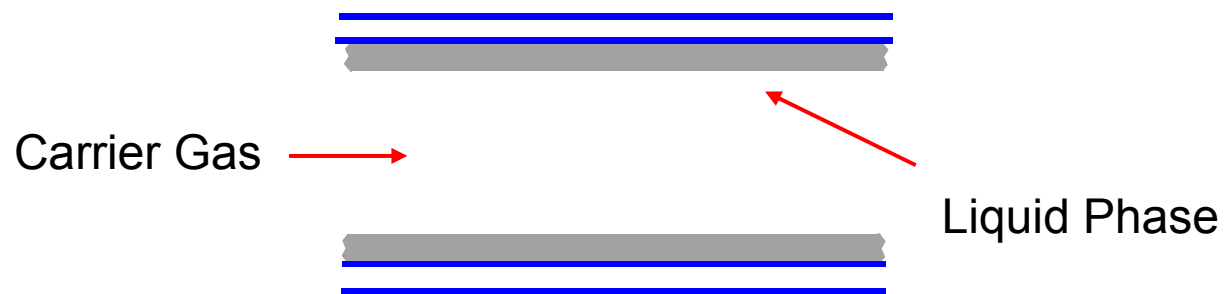


# CAPILLARY COLUMN TYPES

## Porous Layer Open Tube (PLOT)



## Wall Coated Open Tube (WCOT)



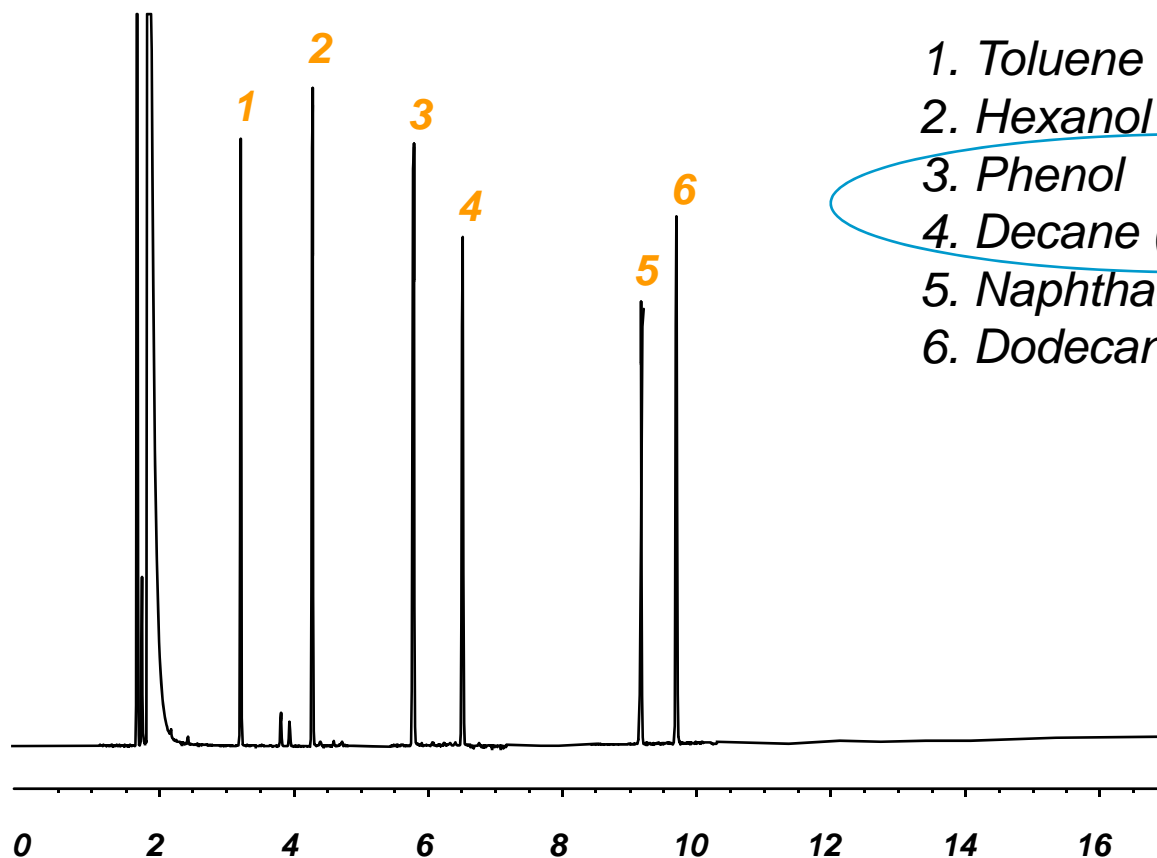
# WCOT Column Types

Agilent J&W has over 50 different stationary phase offerings

Low Polarity			Mid Polarity			High Polarity		
CP-Sil 2	DB & HP-1ms UI	DB & HP-5ms UI	DB-XLB	DB-225ms	DB-ALC1	HP-88	DB-WAX	CP-TCEP
DB-MTBE	DB & HP-1ms	DB & HP-5ms	VF-Xms	DB-225	DB-Dioxin	CP-Sil 88	DB-WAXetr	
CP-Select CB MTBE	VF-1 ms	VF-5ms	DB-35ms UI	CP-Sil 43 CB	DB-200	DB-23	HP-INNOWax	
	DB & HP-1	DB & HP-5	DB & VF-35ms	VF-1701 ms	VF-200ms	VF-23 ms	VF-WAXms	
	CP-Sil 5 CB	CP-Sil 8 CB	DB & HP-35	DB-1701	DB-210		CP-Wax 57 CB	
	Ultra 1	Ultra 2	DB & VF-17ms	CP-Sil 19 CB	DX-4		DB & HP-FFAP	
	DB-1ht	VF-DA	DB-17	HP-Blood Alcohol			DB-WAX FF	
	DB-2887	DB-5.625	HP-50+	DB-ALC2			CP-FFAP CB	
	DB-Petro/ PONA	DB & VF-5ht	DB-17ht	DX-1			CP-WAX 58 FFAP CB	
	CP-Sil PONA CB	CP-Sil PAH CB	DB-608				CP-WAX 52 CB	
	DB-HT SimDis	Select Biodiesel	DB-TPH				CP-WAX 51	
	CP-SimDis	SE-54	DB-502.2				CP-Carbowax 400	
	CP-Volamine		HP-VOC				Carbowax 20M	
	Select Mineral Oil		DB-VRX				HP-20M	
	HP-101		DB-624				CAM	
	SE-30		VF-624ms					
			CP-Select 624 CB					
			DB-1301					
			VF-1301ms					
			CP-Sil 13 CB					



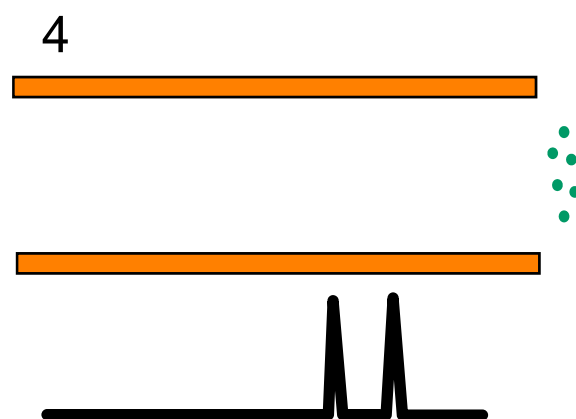
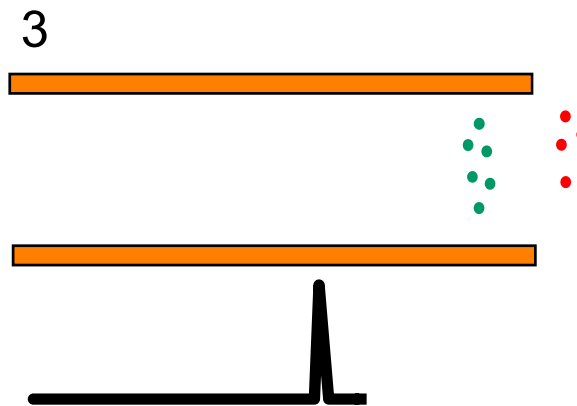
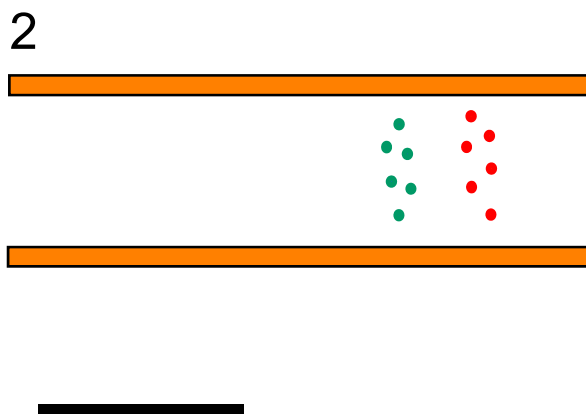
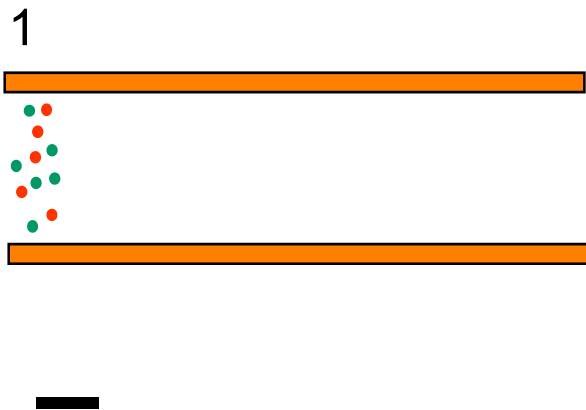
# 100% Methyl Polysiloxane “boiling point column”



- |                   |       |
|-------------------|-------|
| 1. Toluene        | 110°C |
| 2. Hexanol        | 158°C |
| 3. Phenol         | 181°C |
| 4. Decane (C10)   | 174°C |
| 5. Naphthalene    | 218°C |
| 6. Dodecane (C12) | 216°C |

Strong Dispersion  
No Dipole  
No H Bonding

# SEPARATION PROCESS



# TWO PHASES



Solute molecules distribute into the two phases

# DISTRIBUTION CONSTANT ( $K_C$ )



$$N_C^@ = \frac{\text{conc. of solute in stationary phase}}{\text{conc. of solute in mobile phase}}$$

$K_C$  formerly written as  $K_D$

# SOLUTE LOCATION

In stationary phase = Not moving down the column

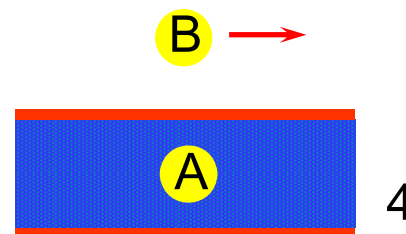
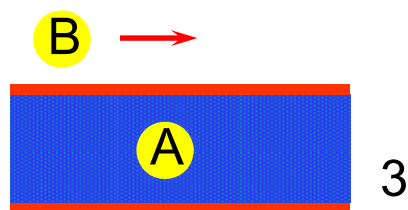
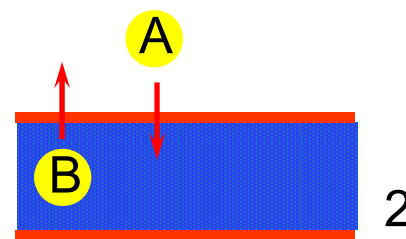
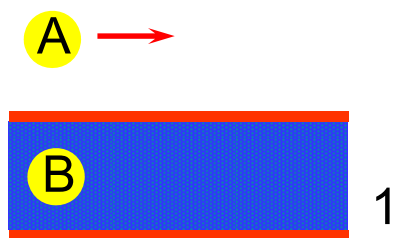
In mobile phase = Moving down the column

# SEPARATION PROCESS

## Movement Down the Column

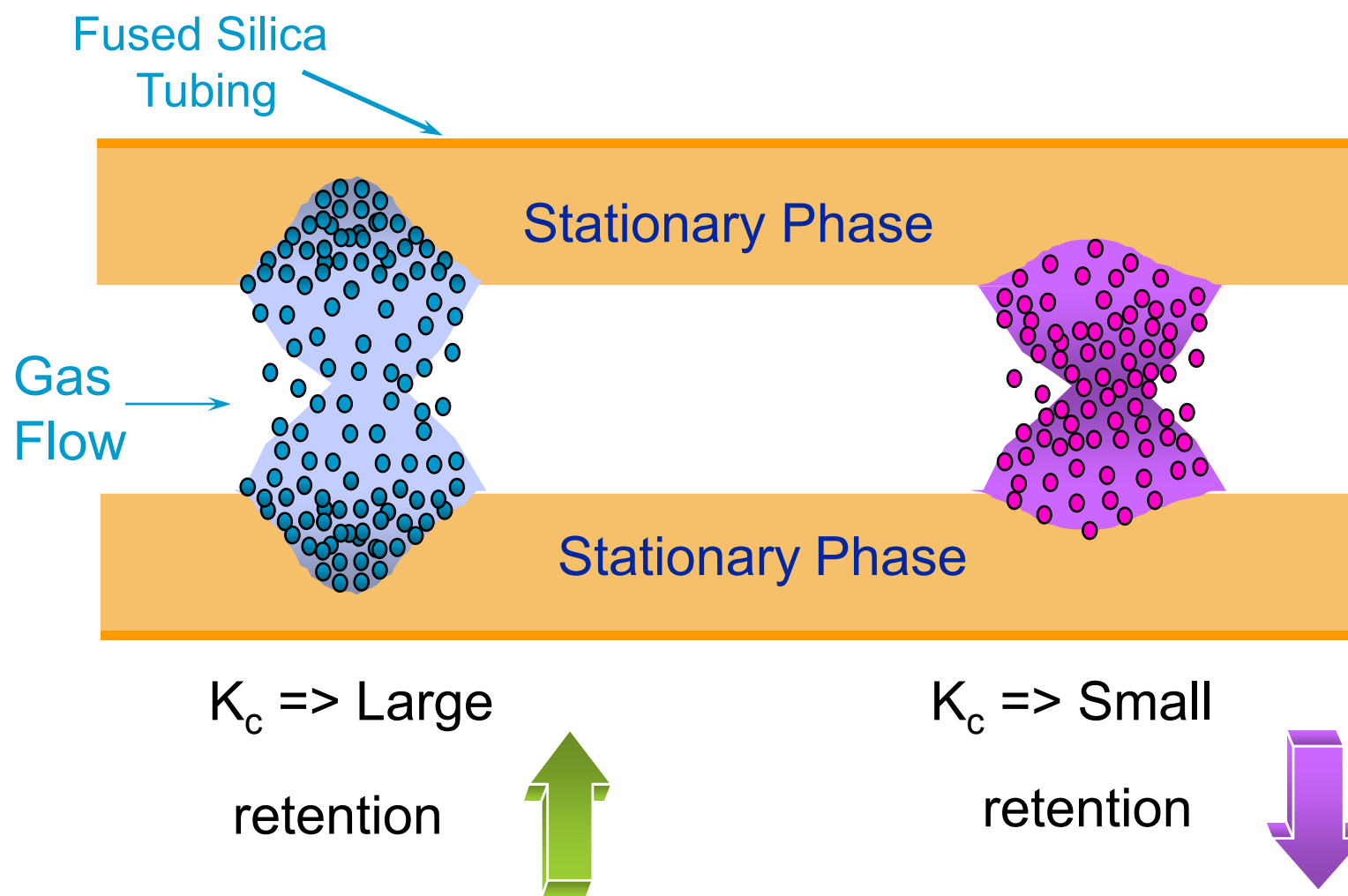
Mobile phase

Stationary phase



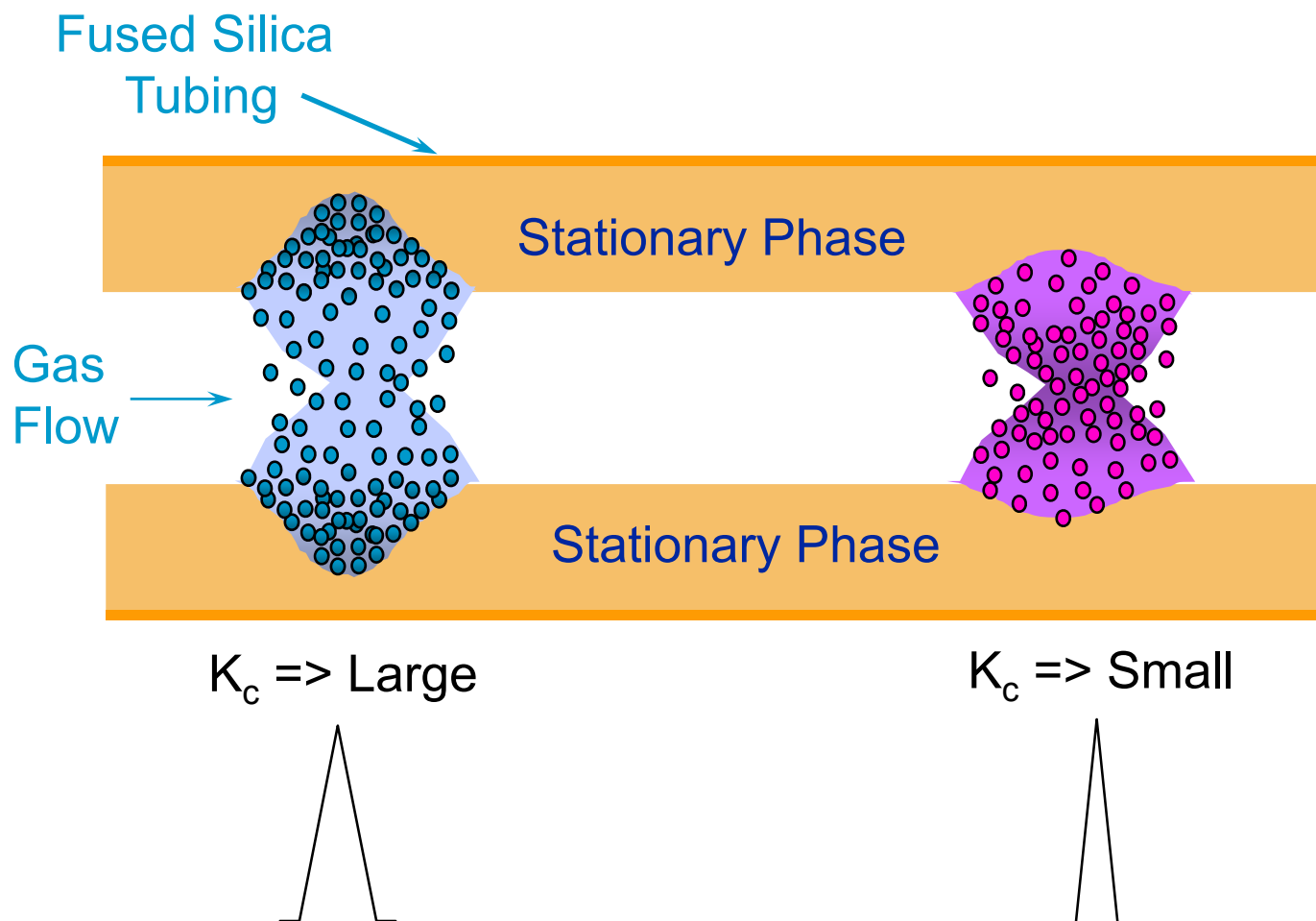


# $K_C$ AND RETENTION



# $K_C$ AND PEAK WIDTH

## Time of Elution



# THREE PARAMETERS THAT AFFECT $K_C$

Solute:

different solubilities in a stationary phase

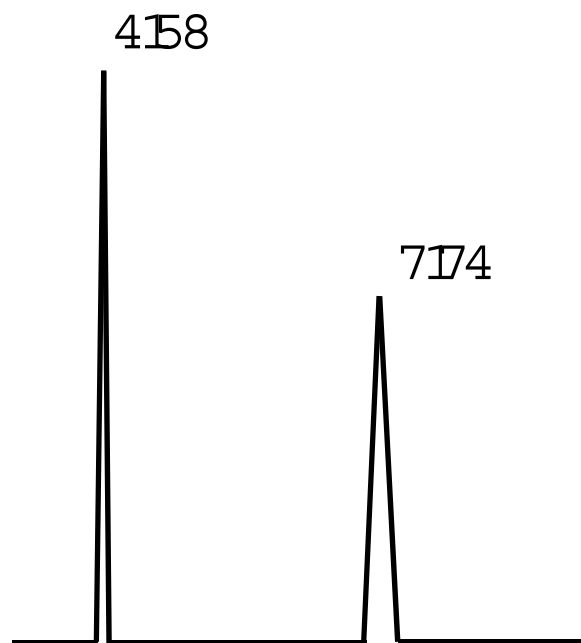
Stationary phase:

different solubilities of a solute

Temperature:

$K_C$  decreases as temperature increases

# RETENTION TIME $t_r$



Time for a solute to travel through the column

# ADJUSTED RETENTION TIME

$t_r'$

Actual time the solute spends in the stationary phase

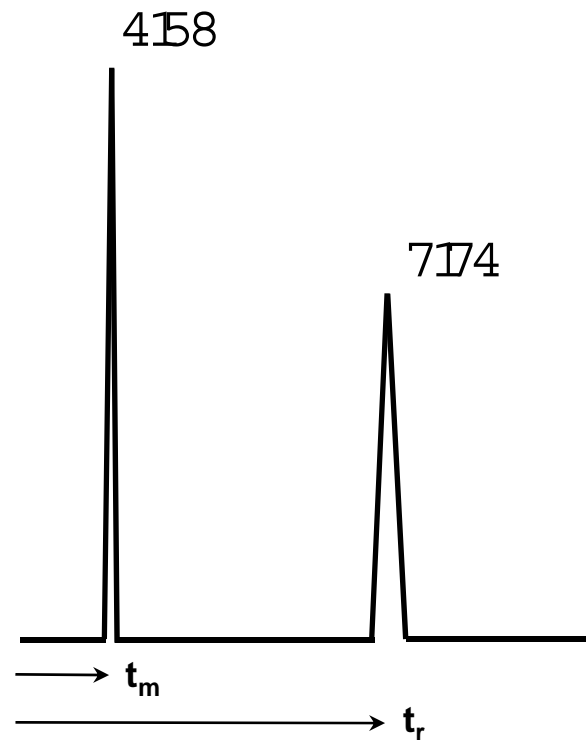
$$t_r' = t_r - t_m$$

$t_r$  = retention time

$t_m$  = retention time of a non-retained solute

# ADJUSTED RETENTION TIME

$t'_r$



$$t'_r = t_r - t_m$$

$$t'_r = 4.41 - 1.25$$

$$t'_r = 3.16 \text{ min} = \text{time spent in stationary phase}$$

# TIME IN THE MOBILE PHASE

*All solutes spend the same amount of time in the mobile phase.*

# RETENTION FACTOR

(k)

Ratio of the time the solute spends in the stationary and mobile phases

$$k' = \frac{t_r - t_m}{t_m}$$

$t_r$  = retention time

$t_m$  = retention time of non-retained compound

Formerly called partition ratio;  $k'$



# RETENTION FACTOR (k)

Relative retention

Linear

Factors out carrier gas influence

# PHASE RATIO ( $\beta$ )

$$\beta = \frac{u}{5g_i}$$

$r$  = radius ( $\mu\text{m}$ )

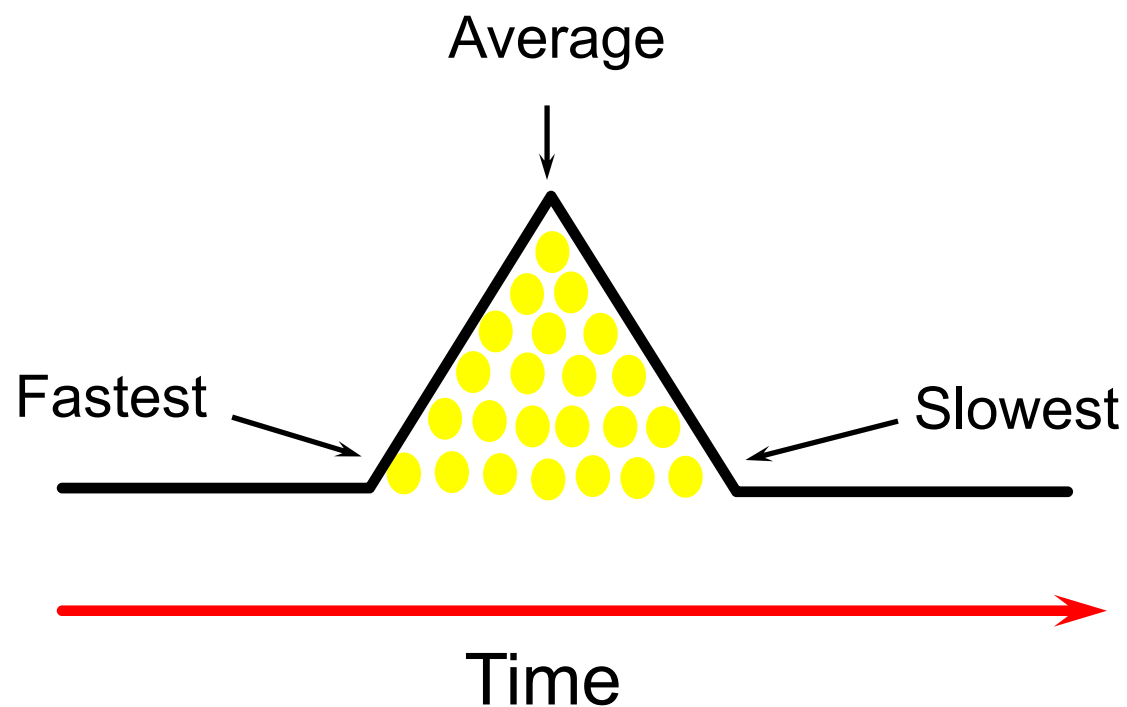
$d_f$  = film thickness ( $\mu\text{m}$ )

# DISTRIBUTION CONSTANT ( $K_c$ )

$$K_c = k\beta$$

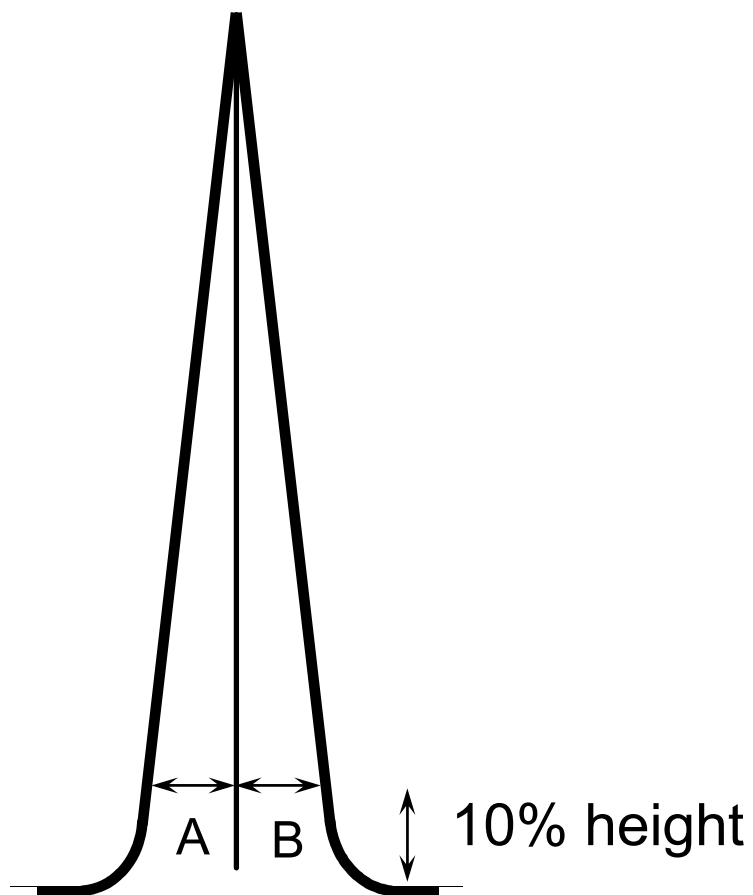
$$n @ \frac{t_r'}{t_m} \quad \beta @ \frac{r}{2d_f}$$

# RANGE OF RETENTION



# PEAK SYMMETRY

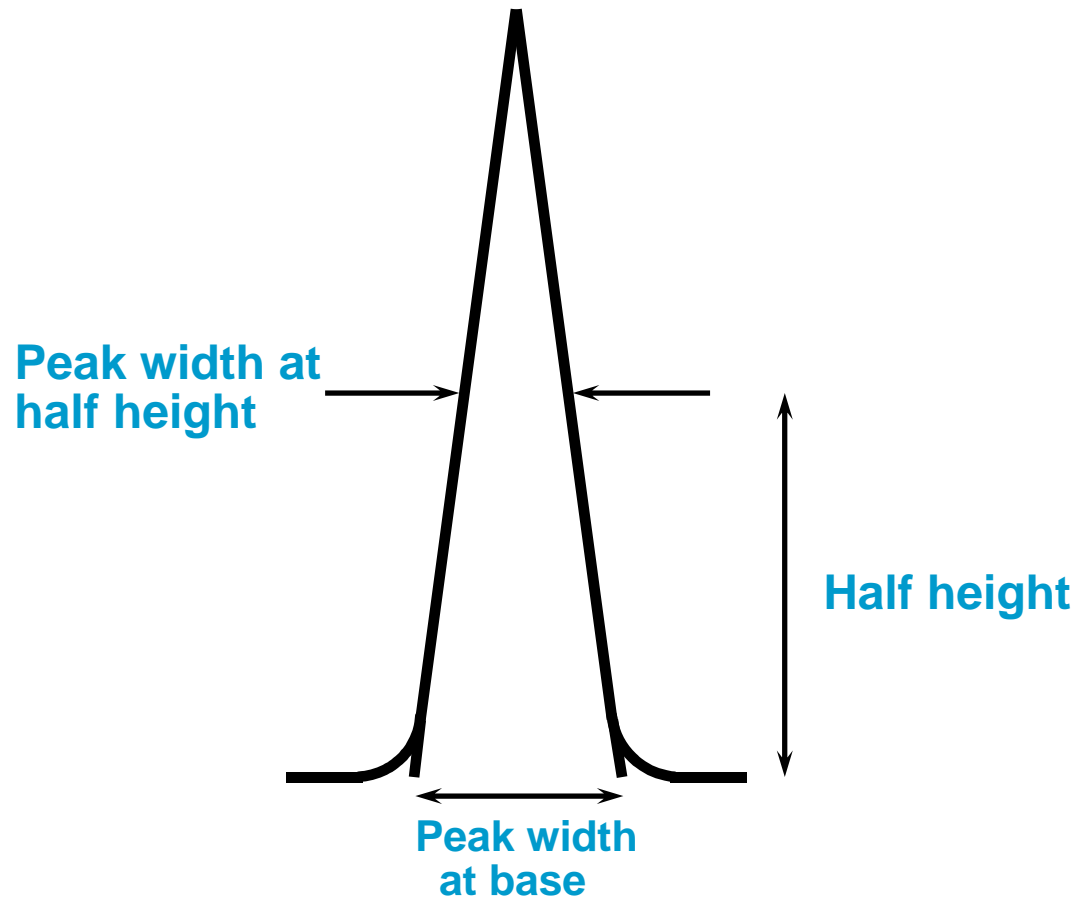
$$\text{Symmetry} = \frac{A}{B}$$



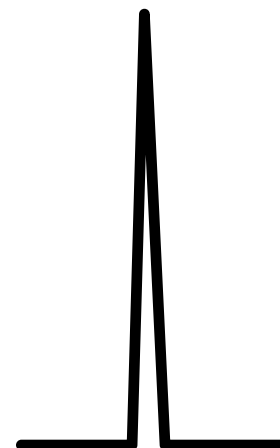
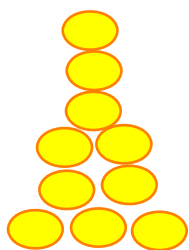
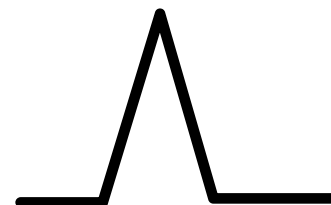
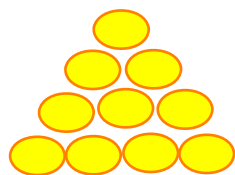
Tailing : Symmetry < 1

Fronting : Symmetry > 1

# PEAK WIDTH



# PEAK WIDTH



# EFFICIENCY

## Theoretical Plates (N)

Large number implies a better column

Often a measure of column quality

Relationship between retention time  
and width



# THEORETICAL PLATES (N)

$$N = 5.545 \left( \frac{t_r}{W_h} \right)^2$$

$t_r$  = retention time

$W_h$  = peak width at half height (time)

# EFFICIENCY MEASUREMENT

## Cautions

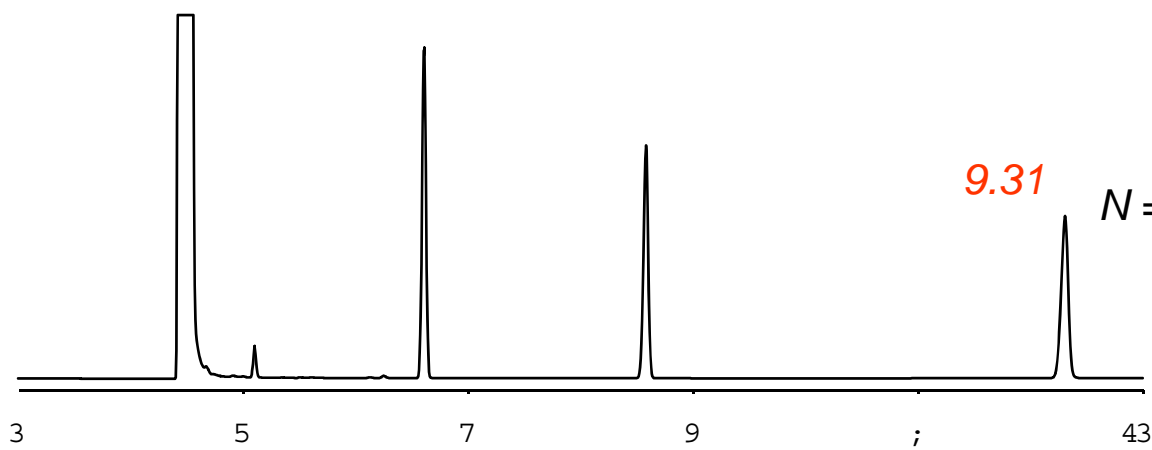
Actually, measurement of the GC system

Condition dependent

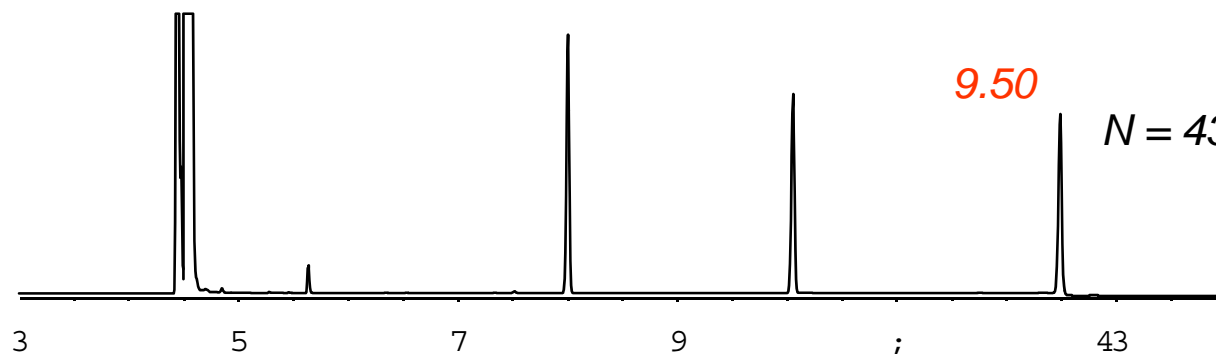
Use a peak with  $k > 5$

# ISOTHERMAL VS. TEMPERATURE PROGRAMMING

## Efficiency



*100°C isothermal*



*75-135°C at 5°/min*

DB-1, 30 m x 0.25 mm ID, 0.25  $\mu$ m  
He at 37 cm/sec  
C10, C11, C12

# SEPARATION VS. RESOLUTION

Separation: time between peaks

Resolution: time between the peaks  
while considering peak  
widths

# SEPARATION FACTOR ( $\alpha$ )

$$\alpha = \frac{k_2}{k_1}$$

co-elution:  $\alpha = 1$

$k_2$  = retention factor of 2nd peak

$k_1$  = retention factor of 1st peak

# RESOLUTION ( $R_s$ )

$$R_s = 1.18 \left( \frac{t_{r2} - t_{r1}}{W_{h1} + W_{h2}} \right)$$

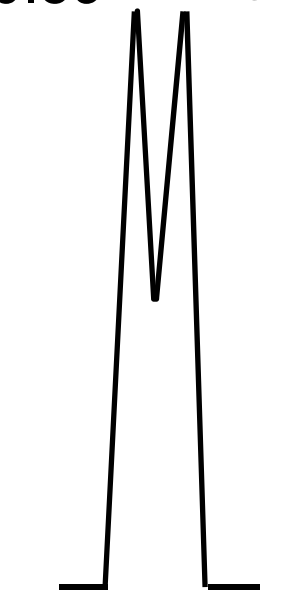
$t_r$  = retention time

$W_h$  = peak width at half height (time)

# RESOLUTION

Baseline Resolution:  $R_s = 1.5$

10.59 10.77

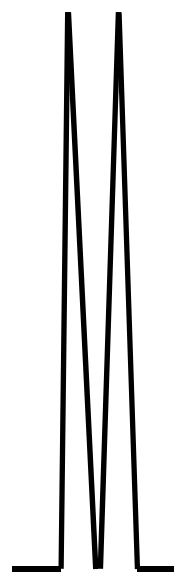


$W_h = 0.105$

$R = 0.84$

$\% = 50$

10.59 10.77

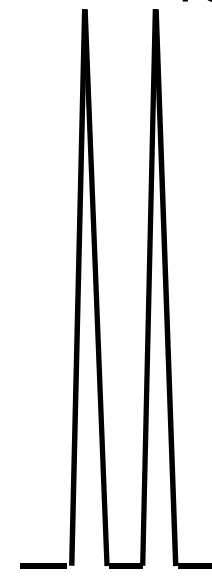


$W_h = 0.059$

$R = 1.50$

$\% = 100$

10.59 10.83



$W_h = 0.059$

$R = 2.40$

$\% = 100$

# Resolution

$$R_s = \frac{\sqrt{N}}{4} \left( \frac{k}{k+1} \right) \left( \frac{\alpha-1}{\alpha} \right)$$

N = Theoretical plates

k = Retention factor

$\alpha$  = Separation factor



# INFLUENCING RESOLUTION

## Variables:

N: column dimensions, carrier gas

a: stationary phase, temperature

k: stationary phase, temperature,  
column dimensions

# Conclusions

The GC is comprised of an inlet, column and detector that all work together to produce good chromatography

Separation (via  $K_C$ ) is based on 3 things:

- Solute: different solubilities/interaction in a given stationary phase
- Stationary phase: different solubilities/interaction of a solute (correct column selection is critical!)
- Temperature:  $K_C$  decreases as temperature increases

When in doubt, contact Agilent Technical Support!

# Additional Recorded – Seminars

<http://www.chem.agilent.com/en-US/Training-Events/eSeminars/14736A/Pages/default.aspx>

[Advanced Topic: Trace Level Analysis for Active Compounds Made Routine with Agilent J&W Ultra Inert Capillary GC Columns](#)

[Advanced Topic – Tips and Tricks of Injector Maintenance](#)

[Advanced Topic – Practical, Faster GC Applications](#)

[Carrier Gases in Capillary GC](#)

[Selection of a Capillary GC Column](#)

[Secrets of GC Column Dimensions](#)

[Techniques for Making Your GC Analysis More Repeatable and Robust](#)

[Techniques, Tips and Tricks of Troubleshooting GC Capillary Systems](#)

[Practical Steps in GC Method Development](#)

[Understanding the Inlets - How to Choose the Right One](#)

# Agilent J&W Scientific Technical Support

**800-227-9770 (phone: US & Canada)\***

**\* *Select option 3..3..1***

**866-422-5571 (fax)**

**GC-Column-Support@agilent.com**

**www.chem.agilent.com**

