Onsite quantitative FTIR analysis of water in turbine oil

Application Note

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Introduction

The availability of the Agilent 5500t FTIR spectrometers, which are compact, easy-to-use and affordable systems, provides new capabilities for real-time, on-site analysis of high value assets such as turbines. With the 5500t FTIR spectrometers, the lubrication specialist now has the ability to monitor key parameters such as oxidation, additive depletion and levels of water in lubricants. In this application brief, we will demonstrate that the Agilent 5500t FTIR spectrometer has the sensitivity, accuracy and reproducibility to determine the level of water in turbine oils without the difficulties associated with the conventional Karl Fischer technique.
Water in turbine oil

An important parameter to measure
The amount of water in turbine oil is critical to the performance and longevity of the equipment. Excessive amounts of entrained water in the turbine oil can cause premature failure of the turbine unit, typically due to changes in the physical properties induced by the presence of water. Physical properties of oil affected by the presence of water include viscosity (measure of the oil’s resistance to flow), specific gravity (density of the oil relative to that of water), and the surface tension (a measure of the stickiness between surface molecules of a liquid). All of these properties are important for the ability of the oil to coat, lubricate, and protect the critical mechanical clearances. In addition, water in turbine oil can accelerate additive depletion and contribute to chemical degradation mechanisms such as oxidation, nitrination, and varnish formation.

On-site analysis is highly desirable
The ability to measure water on-site, as soon as possible after drawing the sample, is a substantial benefit in obtaining accurate water level results. Off-site analysis for trace water in oil may be compromised due to variability of water concentration introduced by storage, transportation, or shipment of a sample. Furthermore, turbine oils contain demulsifying additives that cause microscopic water droplets to separate from the oil and concentrate in layers at the bottom and sides of containers. This demulsifying action takes time to occur, and can cause large variations in analytical measurements. Also, oil samples can sometimes pick up or lose water simply depending on the type of sample container used.

Measuring water in turbine oil
Karl Fischer (KF) coulometric titration is typically used to determine the amount of water in turbine oils. KF has some practical drawbacks for on-site analysis including complicated sample preparation, the use of hazardous and expensive chemical reagents, and length of time required to perform the analysis. However, KF analysis is considered the “gold standard” method for analyzing water in oil because it provides accurate and precise answers.

FTIR spectroscopic analysis eliminates many of the concerns associated with measuring water via KF titration. The spectroscopic method, can be performed in far less time than KF analysis, does not require reagents, and when a rugged and easy-to-use FTIR system such as the 5500t instrument is used, provides on-site analysis. KF titrations require about 10-15 minutes to perform, with the instrument properly conditioned and equilibrated overnight. For KF analysis the oil must be carefully weighed on a high precision balance before and after injecting into the titration vessel. Following each analysis it takes another 5-10 minutes to re-equilibrate. By contrast, the FTIR analysis takes about 2 minutes to perform and is immediately ready for the next sample analysis after simply cleaning with a tissue.

This application note will demonstrate that FTIR spectroscopic analysis using the 5500t FTIR is as accurate and precise as the KF method within the analytical range necessary for measuring water in turbine oil. Using the 5500t, we have developed two FTIR methods for water in turbine oil and have calibrated and evaluated them against the KF titration method.

Water in turbine oil - the FTIR method
Used turbine oil (C&C Oil Co.) was homogenized with water and aged overnight at 70 °C to make a very high water standard. This standard was then diluted with various amounts of a used turbine oil mix, which contains oil in-service four months and another more degraded oil with a dark amber color. These dilutions had various amounts of water based on how much “as is” oil was added. The samples were mixed well and allowed to equilibrate for about an hour before they were analyzed by coulometric Karl Fischer titration (Metrohm 756 KF Coulometer) to determine the concentration of water. The samples were run in duplicate by KF before the infrared spectra were
acquired using the 5500t FTIR spectrometer. The water concentrations for the prepared standards ranged from 22–3720 ppm (parts per million). The water IR absorbance measurement for each standard sample was plotted versus the corresponding KF water data to obtain a residual least squares linear regression. The IR spectra were also analyzed using a partial least squares method to develop a regression model for the quantitative predictions of water in oil.

**Calibration results**

The IR analysis and calibration models indicate a very good correlation between the 5500t FTIR measurements and the Karl Fischer water data. Two different methods were developed for the quantitative measurement of water in oil using the 5500t spectrometer. The first is a relatively simple conventional IR absorbance model following Beer’s Law that uses the region of the IR spectrum in which water strongly absorbs, known as the O-H stretch region. The second method uses multiple regions of the IR spectrum with partial least squares (PLS) chemometric modeling to reduce the effects of noise, baseline variance, and other interfering factors.

**Beer’s law model**

In the first method, a peak area absorbance measurement provides a detection limit of about 30 ppm water in oil (Figure 1). The IR spectra of 15 samples with KF water values ranging from 7-270 ppm were used to build a linear calibration curve that follows Beer’s Law (Figure 2). The weakest water absorbance in Figure 1 is new turbine oil with 30 ppm of water (Red) and the strongest water absorbance is shown in blue with a KF water value of 1460 ppm. The calibration plot is shown in Figure 2 with a correlation coefficient of \( R^2 = 0.977 \) and a standard error of validation (SEV) of \( \sim 40 \) ppm (20-270 ppm range). The addition of higher water concentration standards to the calibration improves the correlation coefficient to \( R^2 = 0.996 \).

Therefore, this calibration is optimized for the low levels of water (<500 ppm), but is still quite accurate for predictions of higher water levels above 500 ppm if necessary.

![Beer’s Law Calibration Plot](image)

**Figure 1.** The overlaid IR spectra of turbine oil with the water absorbance region expanded, water values from bottom to top are 30 ppm (red), 80 ppm (dark green), 217 ppm (light green), 533 ppm (red), and 1460 ppm (blue)

![IR Spectra – Water in Turbine Oil](image)

**Figure 2.** The calibration plot of KF water values (ppm) versus peak absorbance area for water in turbine oil using a Beer’s Law peak area method

**PLS model**

The PLS chemometric model uses more sophisticated mathmatics to develop models that are typically more robust and accurate than the conventional Beer’s Law IR absorbance method demonstrated above. Whereas both the PLS and the Beer’s law quantitative methods for water in oil are sufficient for classification into 100 ppm ranges (i.e. <100 ppm, 100-200 ppm, 200-300 ppm, etc.), the PLS method provides the most accurate
KF water prediction values over the whole range of 30–1500 ppm.

In order to develop the PLS method for water in oil, we used 23 standards covering a range from 7–1460 ppm water. We then recorded the IR spectrum and measured the water level by the KF method. The two sets of results were correlated with partial least squares and the predicted versus actual KF values are plotted in Figure 3 and indicate a correlation coefficient of $R^2=0.983$.

Figure 3. The PLS predicted versus actual plot of KF values using Agilent 4500 Series FTIR spectrometer

Predictions

To validate each FTIR method, 15 unknown mixtures were made by mixing used turbine oils with hydrated turbine oils, and running them by KF (in duplicate) and by FTIR (in triplicate). The coulometric KF performance was verified using 100 ppm and 1000 ppm NIST reference standards. It was found that thorough mixing was important to obtain quality data, due to the heterogeneous nature of water in turbine oil.

Environmental and experimental factors caused the KF duplicate measurements to typically vary by 30–60 ppm, measured consecutively in the 100–1000 ppm range. The FTIR water predictions indicated similar variations in replicate measurements of the same sample. The averages of the replicate measurements by KF and FTIR are compared in Table 1. Good agreement with the KF measurements is observed for both FTIR methods, however, the PLS predictions are statistically better in the 100–1500 ppm range. The standard deviation between the averaged PLS predictions and the averaged KF data (0–700 ppm range) are all below 30 ppm, except for one sample (#11). The Beer’s Law method predictions are better in the 0–100 ppm range, and are sufficient to classify the water concentrations into ranges as follows: <100 ppm, 100–200 ppm, 200–500 ppm, and 500+ ppm.

Conclusions

We have shown that the Agilent 5500t FTIR spectrometer is capable of measuring water in oil at the levels that are critical to the reliable operation of the turbine equipment. The ability to measure water in turbine oil by FTIR spectroscopy eliminates the issues associated with KF measurements, including the need for expensive and hazardous consumables, the time required for the KF measurements, the dependency on the skill of the operator and the operating condition of the KF equipment.
As importantly, the ability to measure water levels at-site via FTIR means that the results will be more accurate, more reproducible and obtained far more rapidly than samples that are sent for off-site analysis to a traditional oil analysis lab. We have observed that low ppm levels of water are observed to change on an hourly basis if left open to air – a sample that initially was 200 ppm can have less than 100 ppm if left in an open sample container overnight. This is also true if the sample container is not filled to the top, and water can evaporate into the head space (air) of the jar.

The Agilent 5500t FTIR spectrometer can detect water at the necessary warning levels. The system can alert when water reaches 100 ppm and then issue a critical warning if the water reaches 200 ppm. In addition to the analysis of water, Agilent’s mobile FTIR spectrometers can measure the depletion of additives and determine the levels of oxidation and nitration by-products in turbine oils.

Further advances in water-in-oil measurement, with the Agilent 4500 and 5500 FTIR spectrometers, have been made using a proprietary water stabilizer technique as documented in Agilent Application Note 5991-0672EN: Onsite FTIR quantitative analysis of water in mineral-based oils using a novel water stabilization technique. This more advanced technique features much improved accuracy, reproducibility, and robustness for multiple oil types.